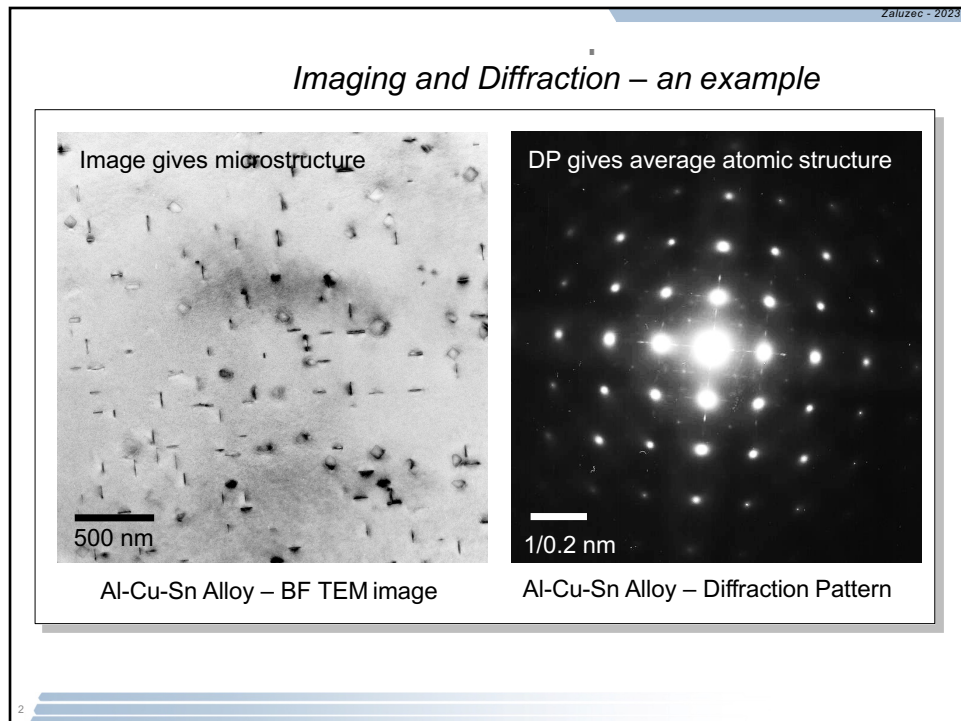
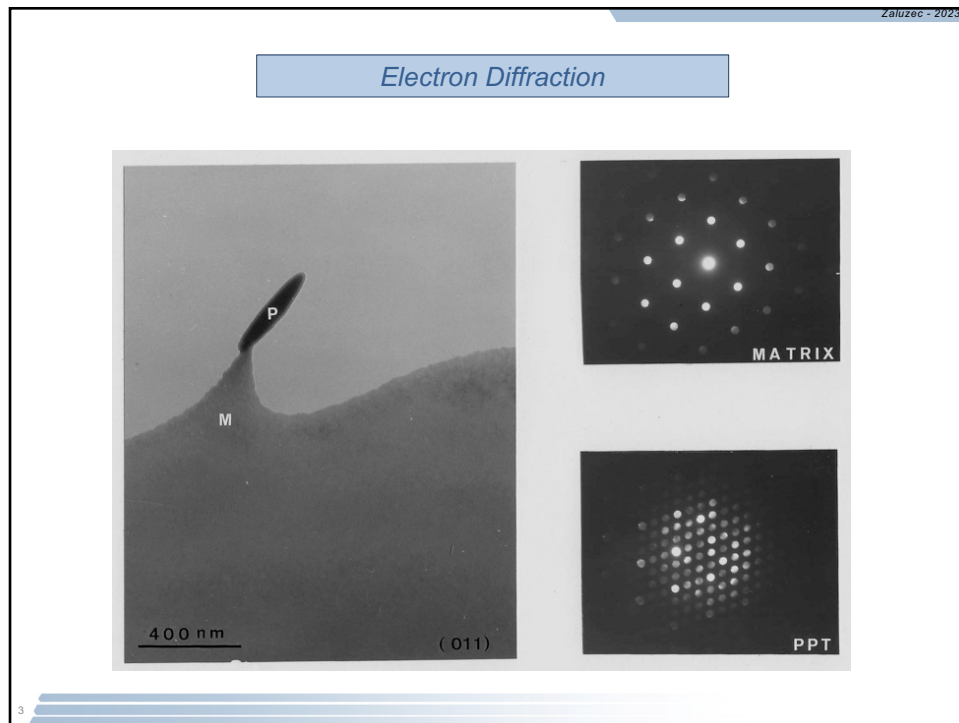


1



2



3

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Roles of the Lenses

Gun Lens

Helps form probe

Condenser Lens

Mainly controls:
Spot Size
hence total beam
current

Objective Lens

Mainly controls
Focus, 1st Magnification

Diffraction/Intermediate Lens

Controls Mode

Projector Lens

Magnification

Most TEM/STEM
have 7-9 Lenses

- 1 Gun Lens
- 2-3 Condensers
- 1 Objective
- 1-2 Intermediate
- 1-2 Projectors

Most instruments
Have Electromagnetic
Round Lenses

Exception:
Aberration Corrected
Systems

Note the locations
of the various
Apertures.


Optimum aperture
sizes are needed
for various
imaging functions.

4

4

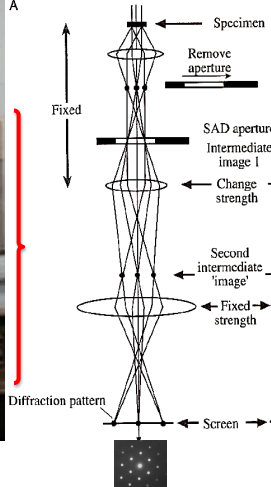
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Projection/Imaging System



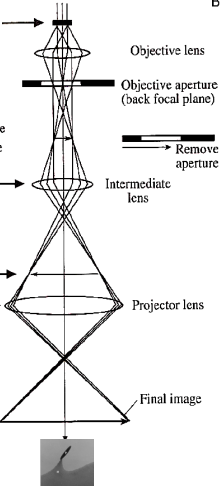
Projects/Magnifies
the Diffraction Pattern
(Back Focal Plane)

A



Projects/Magnifies
the Image
(Image Plane)

B



- The TEM is unique in that we can easily carry out **imaging** and **diffraction** in the **same instrument**

This is carried out by adjusting the strength of the intermediate lenses in the **projection system**

Other instruments can do this but it is much more difficult!

5

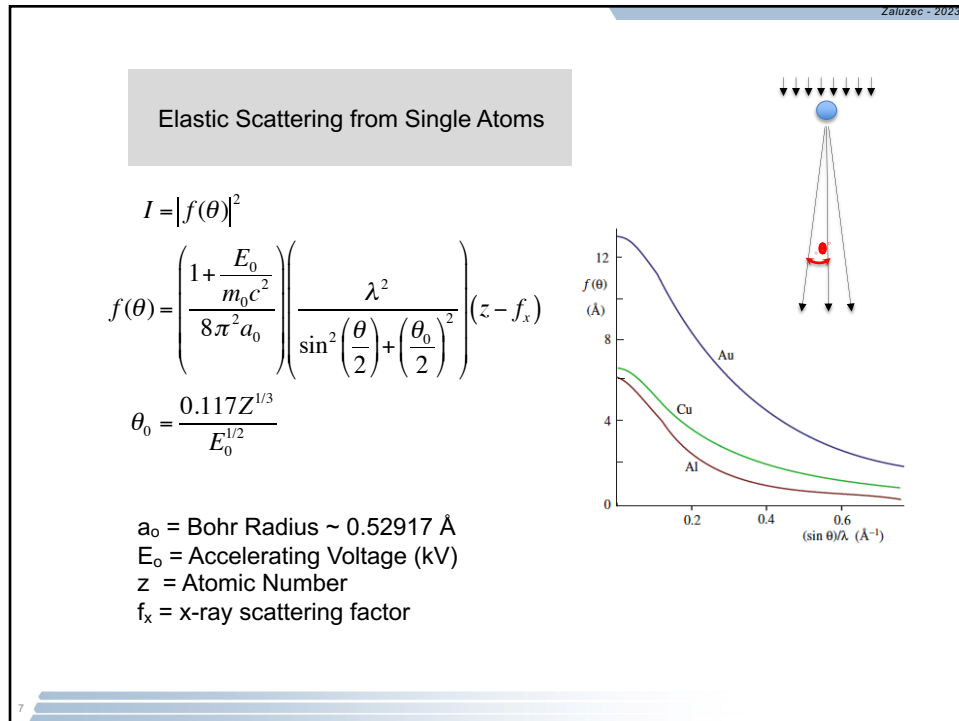
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Electron Diffraction

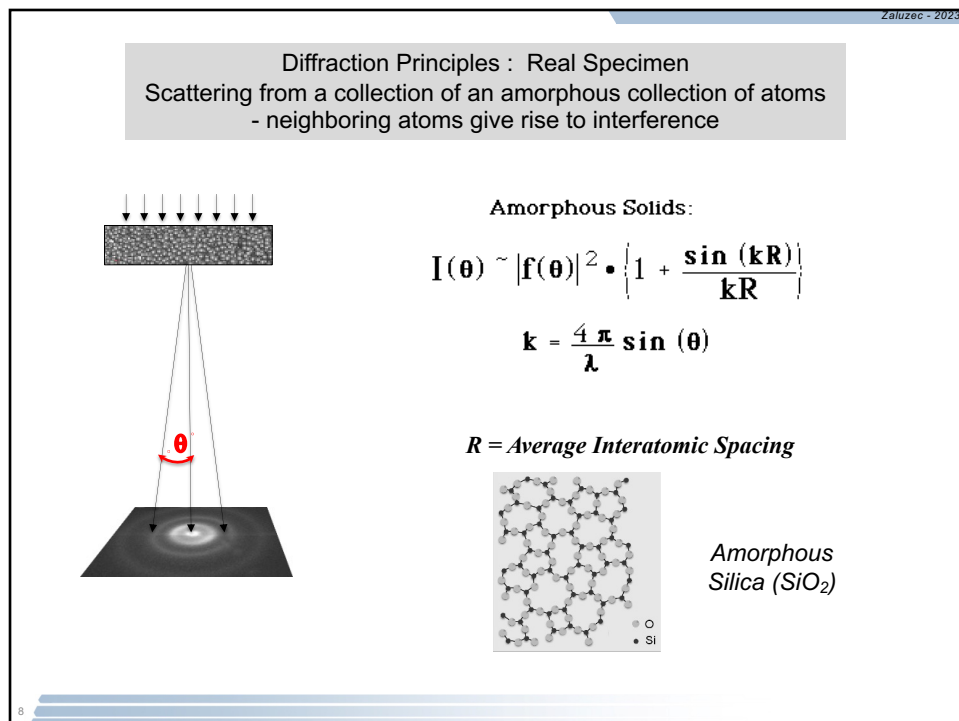
- Diffraction pattern capability is one of the most important features of the TEM, because we can relate the crystallography to the images obtained.
- The ability to determine crystallographic orientations locally (down to the nm level) gives the TEM its great advantage over the SEM and visible-light microscopes.
- The questions that we can address using diffraction patterns obtained in the TEM include the following:
 - Is the specimen crystalline? Crystalline and amorphous materials have very different properties.
 - If it is crystalline, then what are the crystallographic characteristics (lattice parameter, symmetry, etc.) of the specimen?
 - Is the specimen monocrystalline? If not, what is the grain morphology, how large are the grains, what is the grain-size distribution, etc?
 - What is the orientation of the specimen or of individual grains with respect to the electron beam?
 - Is more than one phase present in the specimen?

6

3



7



8

Scattering from an collection of an amorphous collection of atoms
 - neighboring atoms give rise to interference

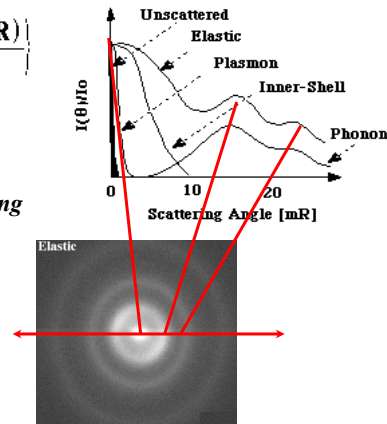
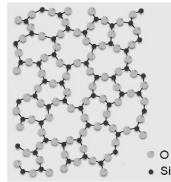
Amorphous Solids:

$$I(\theta) \sim |f(\theta)|^2 \cdot \left\{ 1 + \frac{\sin(kR)}{kR} \right\}$$

$$k = \frac{4\pi}{\lambda} \sin(\theta)$$

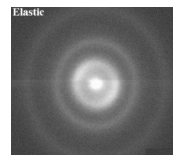
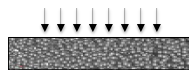
$R = \text{Average Interatomic Spacing}$

Amorphous
Silica (SiO_2)



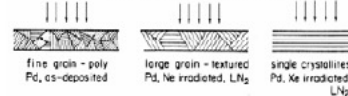
9

Disordered
Amorphous
Non-Crystalline

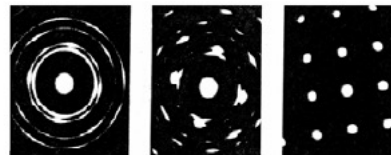
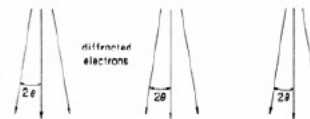


Ordered
Crystalline

200 keV electrons - TEM mode

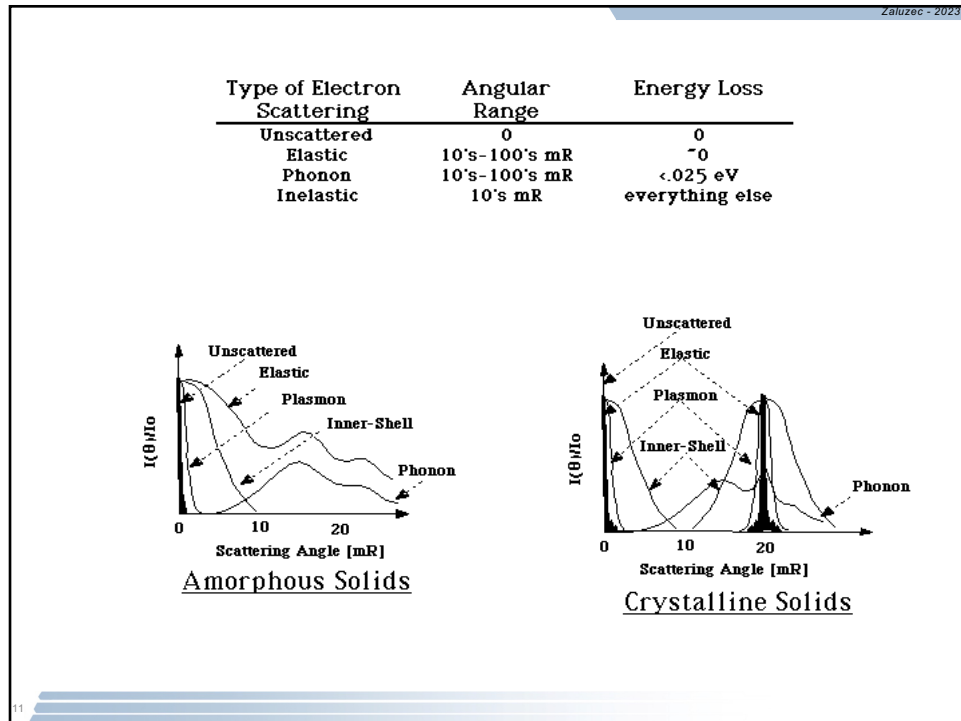


Pd film

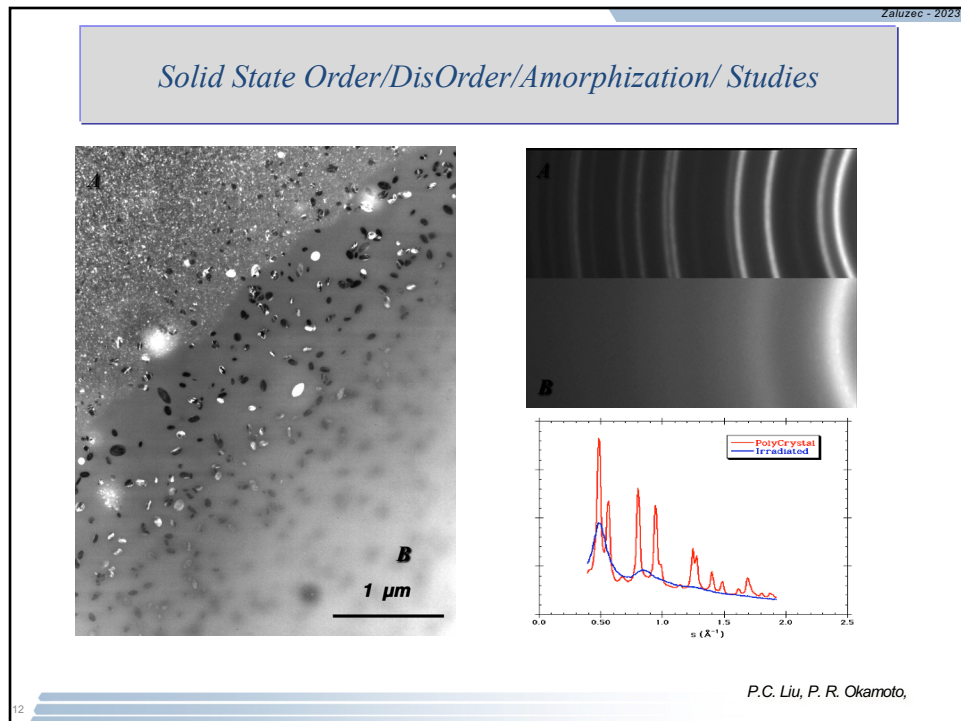


10

10



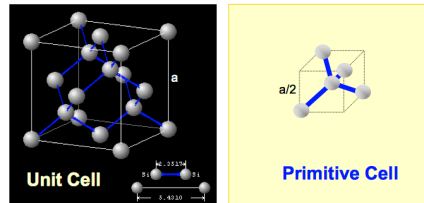
11



12

Crystal Structures

- ❑ The periodic arrangement of the atoms is called the **Lattice**.
- ❑ **Unit Cell:**
Representative of the entire lattice and is regularly repeated throughout the crystal.
- ❑ **Primitive Cell:**
Smallest unit cell which can be repeated to form the lattices.



Each crystal built up of a repetitive stacking of unit cells each identical in size, shape, and orientation with every other one.

13

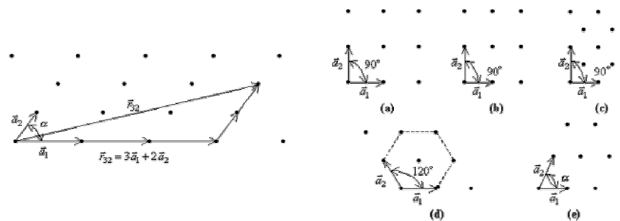
Crystal Structures

- ❑ Coordinates of position in the unit cell
- ❖ x, y, z expressed in terms of the unit cell edges.

$$\vec{r}_{xyz} = x\vec{a} + y\vec{b} + z\vec{c}$$

Example

\vec{r}_{321} reached by moving along the axis a distance of $3x$ the length of the vector \vec{a} , the parallel to \vec{b} , a distance $2 \times \vec{b}$, and finally parallel to \vec{c} , a distance equal to the length of \vec{c} .



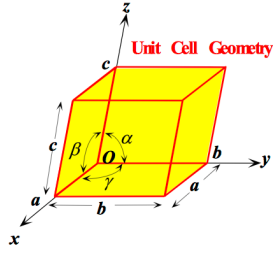
14

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Crystal Lattice Group

☐ Bravais lattices

Length and Angle

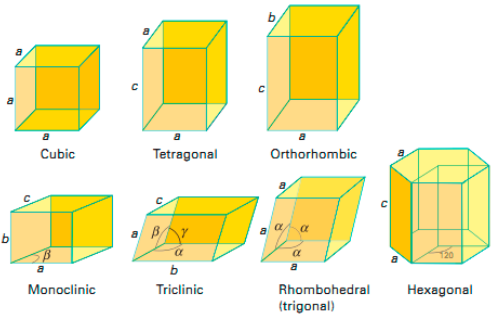


Triclinic	$a \neq b \neq c$	$\alpha \neq \beta \neq \gamma \neq 90^\circ$	K_2CrO_7
Monoclinic	$a \neq b \neq c$	$\alpha = \gamma = 90^\circ \neq \beta$	$\beta\text{-S}, CaSO_4 \cdot 2H_2O$
Orthorhombic	$a \neq b \neq c$	$\alpha = \beta = \gamma = 90^\circ$	$\alpha\text{-S}, Ga, Fe_3C$
Tetragonal	$a = b \neq c$	$\alpha = \beta = \gamma = 90^\circ$	$\beta\text{-Sn}, TiO_2$
Cubic	$a = b = c$	$\alpha = \beta = \gamma = 90^\circ$	$Cu, Ag, Zn, NaCl$
Hexagonal	$a_1 = a_2 = a_3 \neq c$	$\alpha = \beta = 90^\circ, \gamma = 120^\circ$	Zn, Cd
Rhombohedral	$a = b = c$	$\alpha = \beta = \gamma \neq 90^\circ$	As, Sb, Bi

15

15

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7 Crystal Systems

14 Bravais Lattices

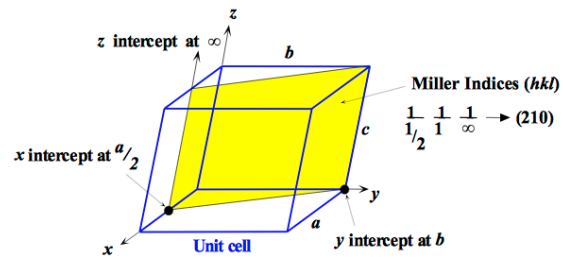
Sr. No.	Crystal System	Axial length of Unit Cell	Inter axial angles	Number of Lattice in the system
1	Cubic	$a = b = c$	$\alpha = \beta = \gamma = 90^\circ$	3
2	Tetragonal	$a = b \neq c$	$\alpha = \beta = \gamma = 90^\circ$	2
3	Orthorhombic	$a \neq b \neq c$	$\alpha = \beta = \gamma = 90^\circ$	4
4	Monoclinic	$a \neq b \neq c$	$\alpha = \beta = 90^\circ \neq \gamma$	2
5	Triclinic	$a \neq b \neq c$	$\alpha \neq \beta \neq \gamma \neq 90^\circ$	1
6	Trigonal	$a = b = c$	$\alpha = \beta = \gamma < 120^\circ, \neq 90^\circ$	1
7	Hexagonal	$a = b \neq c$	$\alpha = \beta = 90^\circ, \text{ and } \gamma = 120^\circ$	1

16

16

Crystal Plans

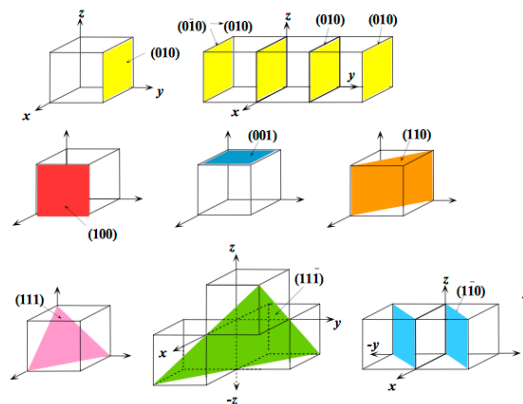
□ Identification of a plan in a crystal



17

Crystal Planes in the Cubic Lattice

□ Various plans in cubic lattice



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Miller Convention Summary

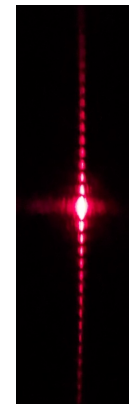
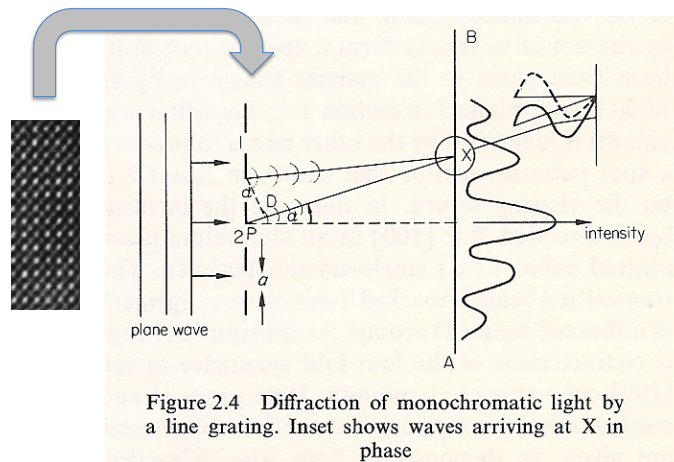
Convention	Interpretation
(hkl)	Crystal Plane
$\{hkl\}$	Equivalent Planes
$[hkl]$	Crystal Direction
$\langle hkl \rangle$	Equivalent Directions

Examples

- plane $\{111\}$: (111) (-111) $(1-11)$ $(11-1)$
- direction $\langle 111 \rangle$: $[111]$ $[-111]$ $[1-11]$ $[11-1]$

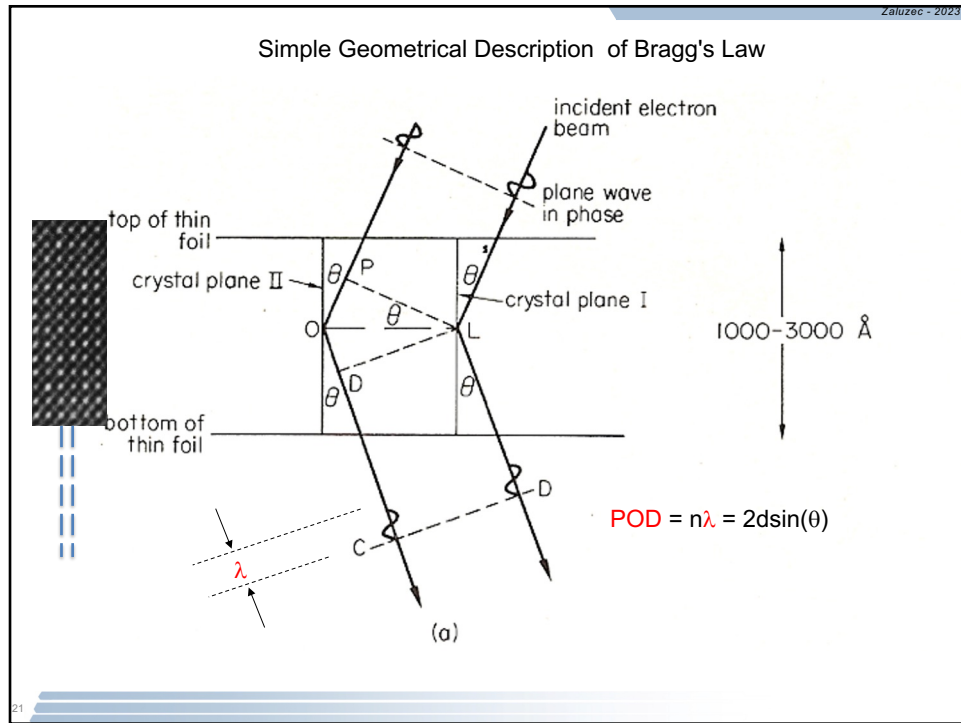
19

Periodic Structures Create Constructive/Destructive Interference

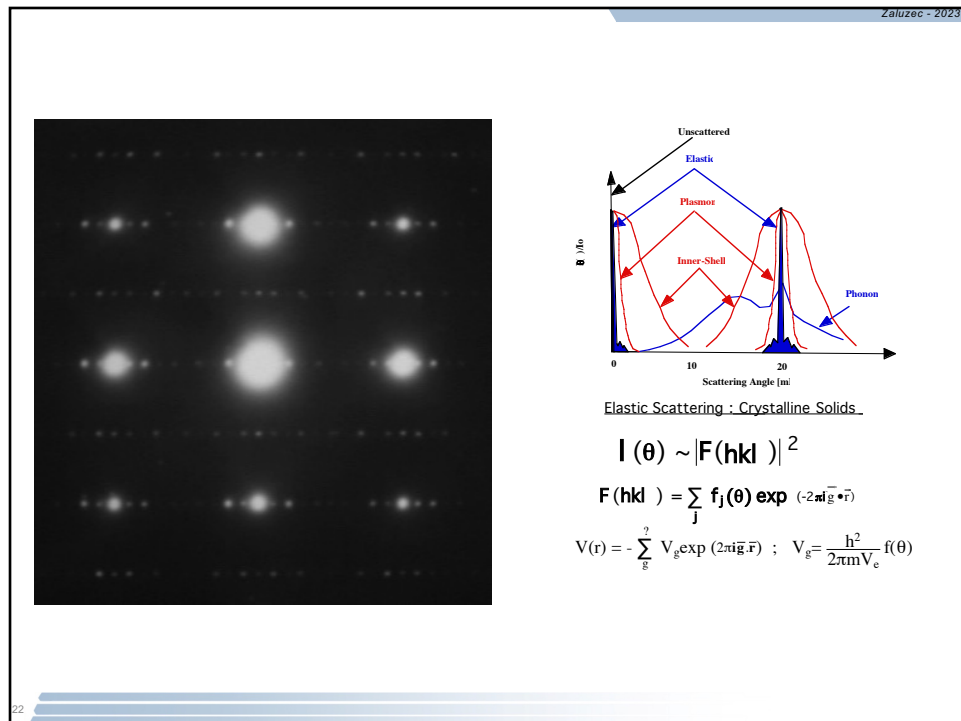


Young's Slit
interference
Experiment

20



21



22

Type of Diffraction Patterns

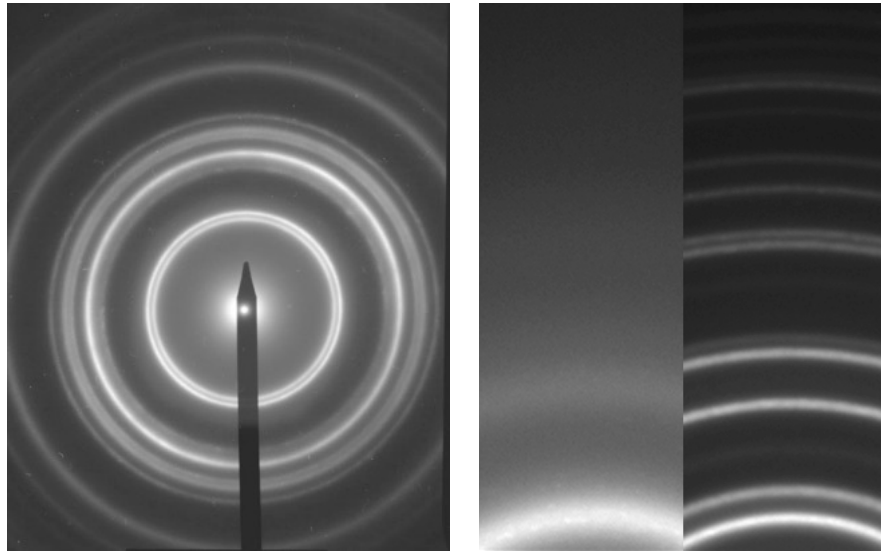
- Electron diffraction patterns produced in transmission in the electron microscope can be of three different types.
 - (a) Ring pattern Polycrystalline, and amorphous specimen
 - (b) Spot pattern Single-crystal region of the specimen
 - (c) Kikuchi line pattern
- Type (b) and (c) often occur on the same diffraction pattern. In general, spot and Kikuchi patterns will be taken from a specific area of the specimen and are known as 'selected area' diffraction patterns (SADPs).

Ring Pattern:

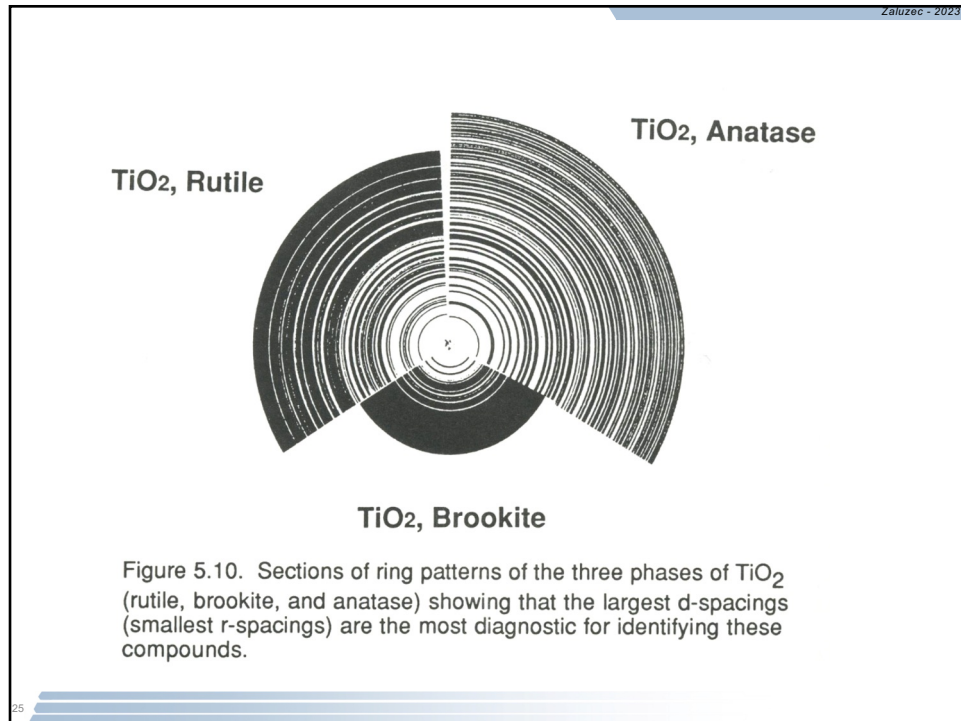
- The major use of ring patterns is in the identification of phases using extraction replicas. These diffraction patterns also arise from very fine grain size polycrystalline material such as physically or chemically vapor deposited or electrodeposited thin foils.

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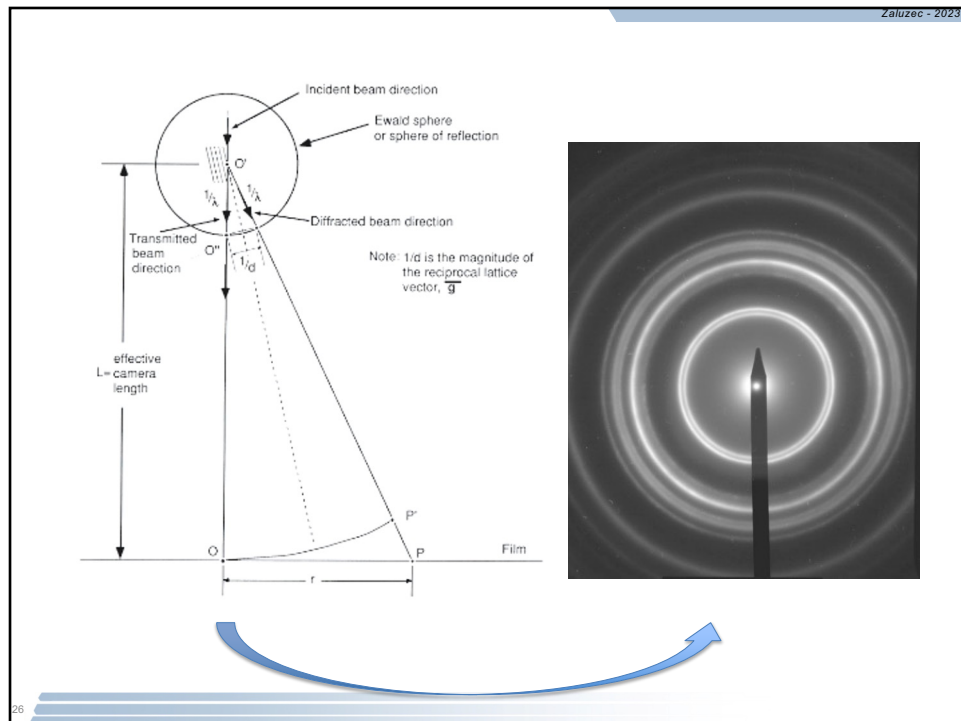
Polycrystalline vs Disordered/Amorphous



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25



26

Ring pattern: what can we obtain

- d-spacing

$$Rd_{hkl} = L\lambda$$

R: the measured ring radius

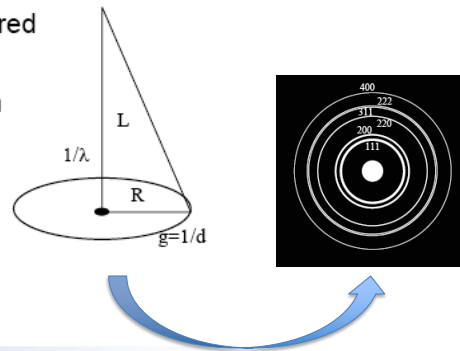
d_{hkl} : the d-spacing being measured

L: camera length

λ : wave length of electron beam

- Camera length calibration

- Crystalline / particle fineness



27

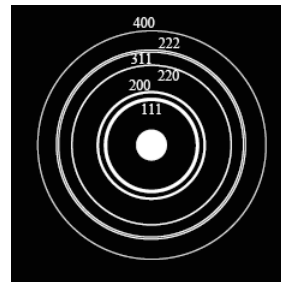
Example

Camera Length Calibration
Aluminum Polycrystalline Specimen

$$Rd_{hkl} = \lambda L$$

$$a_0 = 0.40497 \text{ nm}$$

$$d_{hkl} = \frac{a_0}{\sqrt{h^2 + k^2 + l^2}}$$

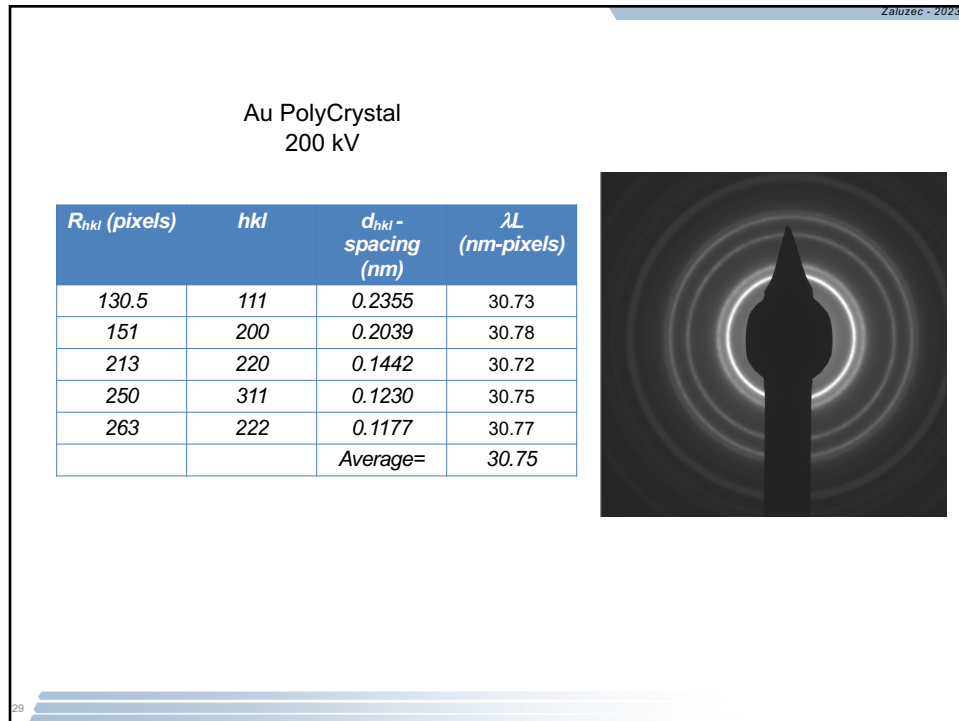


← R →

$R_{hkl} \text{ (mm)}$	hkl	$d_{hkl} \text{ - spacing (nm)}$	$\lambda L \text{ (nm-mm)}$
10.3	111	0.2338	2.409
11.9	200	0.2025	2.409
16.8	220	0.1432	2.406
19.7	311	0.1221	2.405
20.6	222	0.1169	2.408
		Average=	2.407 nm-mm

28

28



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Indexing Diffraction patterns

The Ring Patterns

The radius of each ring is characteristic of the spacing of the reflecting planes in the crystal and the magnification settings of the microscope lenses.

Procedure for indexing ring patterns is as follows:

- When the identity of the material is known, we have the following:
 - Measure the ring diameters.
 - Determine the ratios of the squares of the diameters of the outer rings to that of the first or second (low-index) ring.
 - Check the ratios against a table of ratios of the interplanar spacings for the crystal structure of interest, see table 2.2.
- When the identity of the substance is unknown, we have the following:
 - Measure the diameter of the rings.
 - Convert the distances into interplanar spacings using the camera constant defined as $Rd = \lambda L$.
 - Use ASTM (American Society for Testing Materials) index to identify the phase, starting with the most likely on the basis of the known constituents of the alloy.

Diffraction patterns from polycrystalline specimens are most commonly used either to calibrate the camera length or to identify precipitates.

30

30

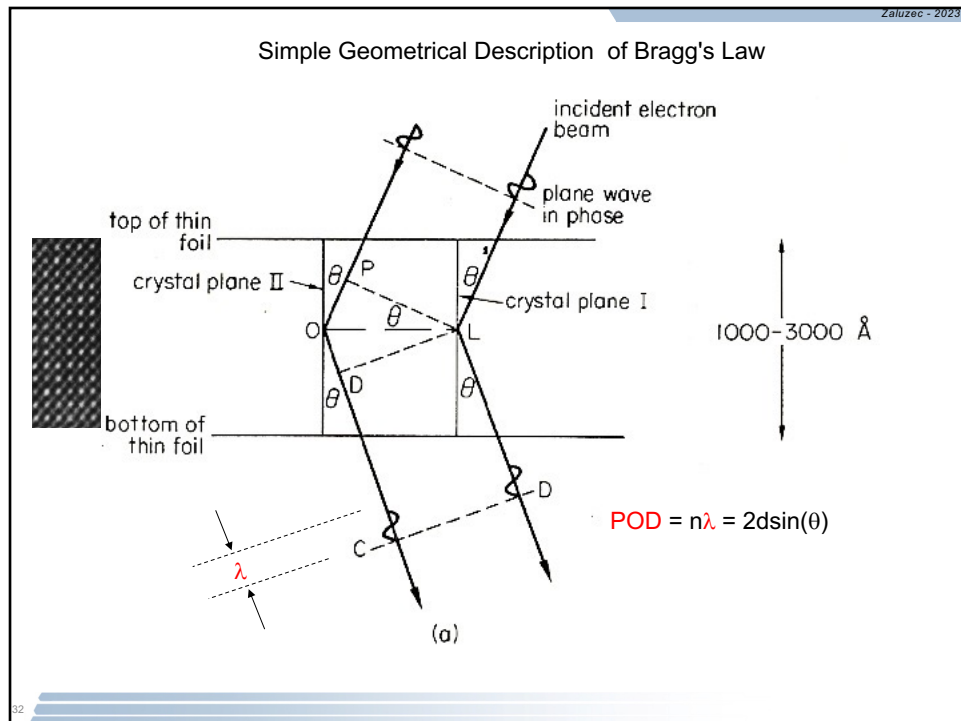
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Table 2.2 Proportionalities of the ratios of the radii of ring patterns for different crystal structures

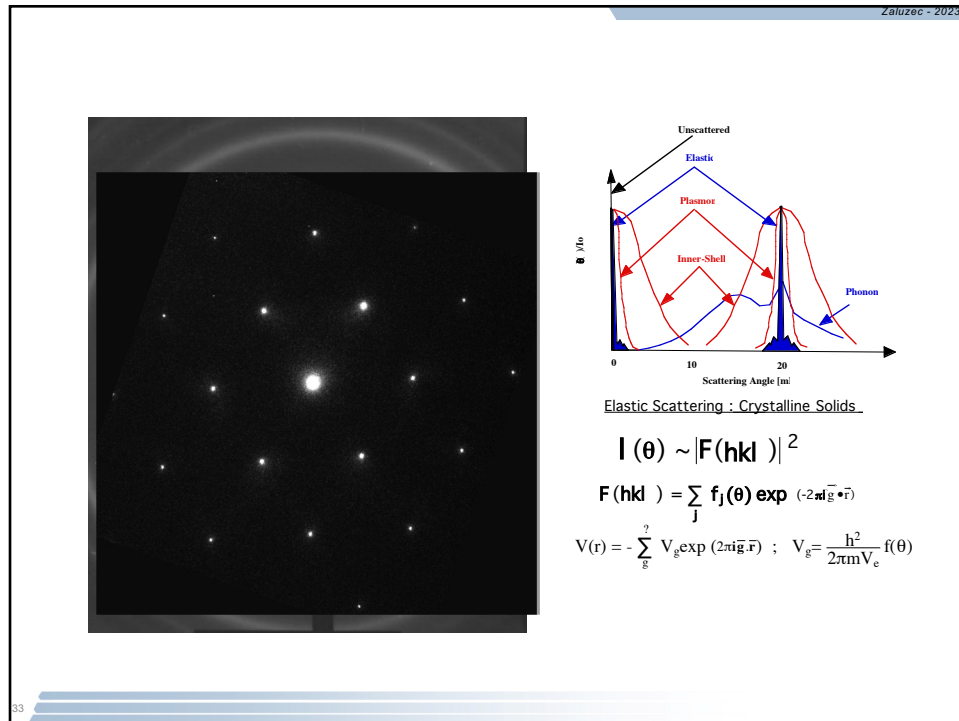
Crystal structure	Formula for interplanar spacing	Possible values of h, k, l for reflection (up to 20)	Criterion
simple cubic	$\frac{1}{d^2} = \frac{h^2 + k^2 + l^2}{a^2} = \frac{N}{a^2}$	N an integer except 7 or 15	ratios of squares of radii $\propto N$
f.c.c.	$\frac{1}{d^2} = \frac{h^2 + k^2 + l^2}{a^2} = \frac{N}{a^2}$	$N = 3, 4, 8, 11, 12, 16, 19, 20$	ratios $\propto N$
b.c.c.	$\frac{1}{d^2} = \frac{h^2 + k^2 + l^2}{a^2} = \frac{N}{a^2}$	$N = 2, 4, 6, 8, 10, 12, 14, 16, 18, 20$	ratios $\propto N$
diamond structure	$\frac{1}{d^2} = \frac{h^2 + k^2 + l^2}{a^2} = \frac{N}{a^2}$	$N = 2, 8, 11, 16, 19$	ratios $\propto N$
tetragonal	$\frac{1}{d^2} = \frac{h^2 + k^2}{a^2} + \frac{l^2}{c^2}$	$h^2 + k^2 = 1, 2, 4, 5, 8, 9, 10, 13, 16, 17, 18, 20$	ratios frequently proportional to 2; use Bunn chart, see Henry <i>et al.</i> (1951)
hexagonal	$\frac{1}{d^2} = \frac{4}{3} \frac{h^2 + hk + k^2}{a^2} + \frac{l^2}{c^2}$	$h^2 + hk + l^2 = 1, 3, 4, 7, 9, 12, 13, 16, 19$	ratios frequently proportional to 3; use Bunn chart, see Henry <i>et al.</i> (1951)

31

31



32



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Index Simple Patterns – spots are produced by planes in one zone

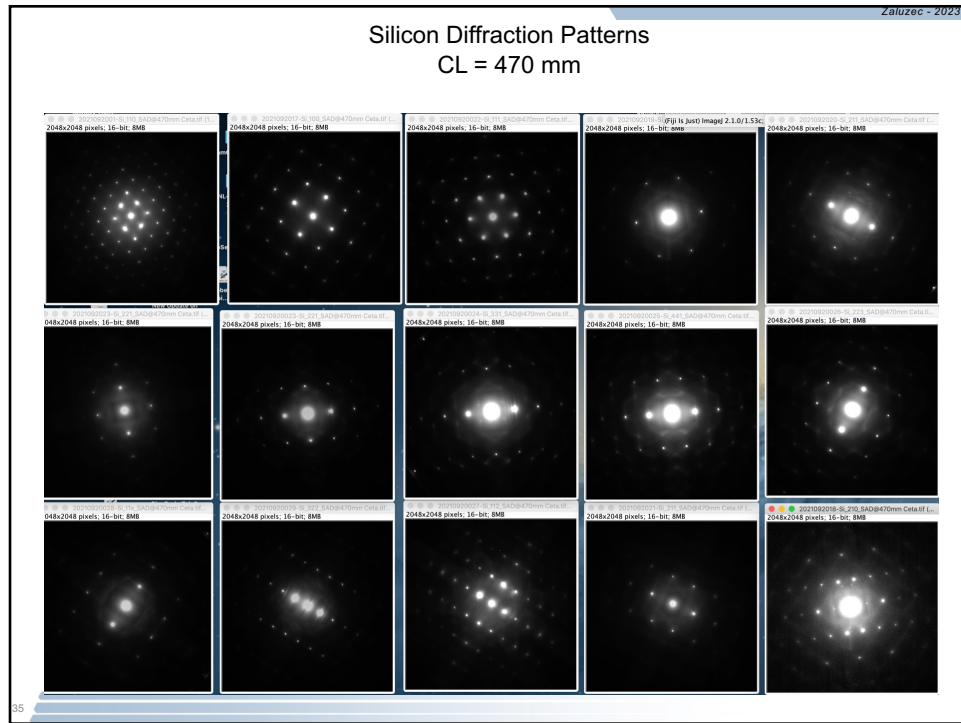
- Frequently these patterns may be easily recognized from their simplicity and their six-, four-, three- and two-fold symmetry.

(a) *Indexing the pattern.* The indexing procedure involves one of two procedures:

- (i) Camera constant method – Camera constant known, materials known. Measure the distance R of the spot from the center spot, figure 2.20 (a) (most accurately determined by measuring the total distance between several spots in this direction and dividing as necessary). Divide the camera constant λL ($Rd = \lambda L$) by R and check the result against a list of d spacings.
- (ii) Method of ratios. Camera constant unknown. Material λL known. Measure distances of spots from center spot relative to that between the nearest spot and the center. Check against tables of ratios of d spacings for low-index planes. Account must be taken of the occurrence of planes with the same d spacing, see table A 4.1.

34

34

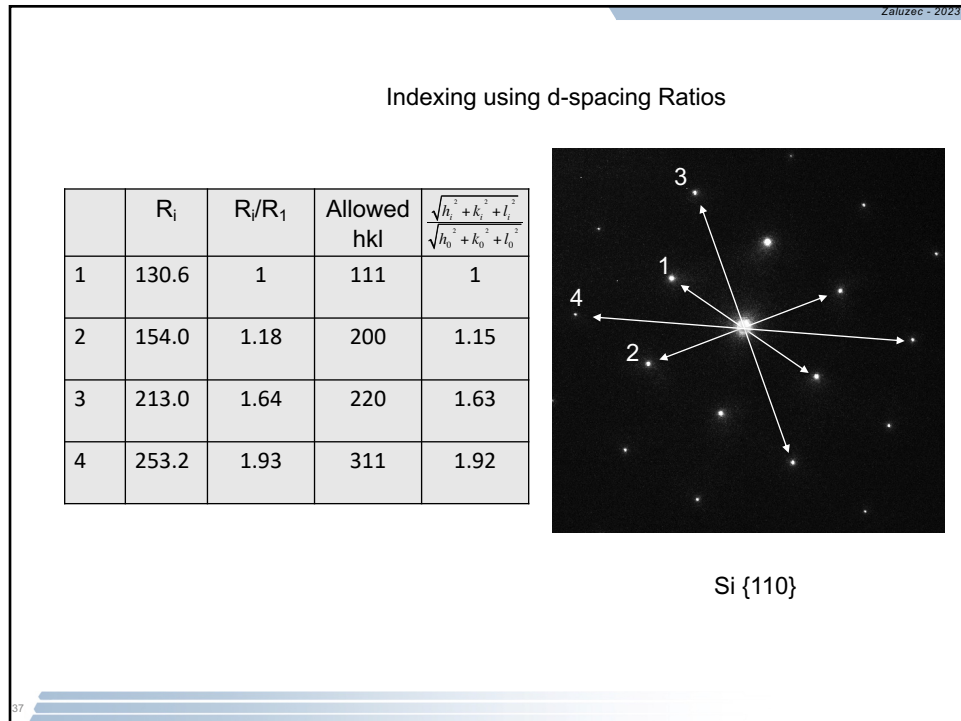


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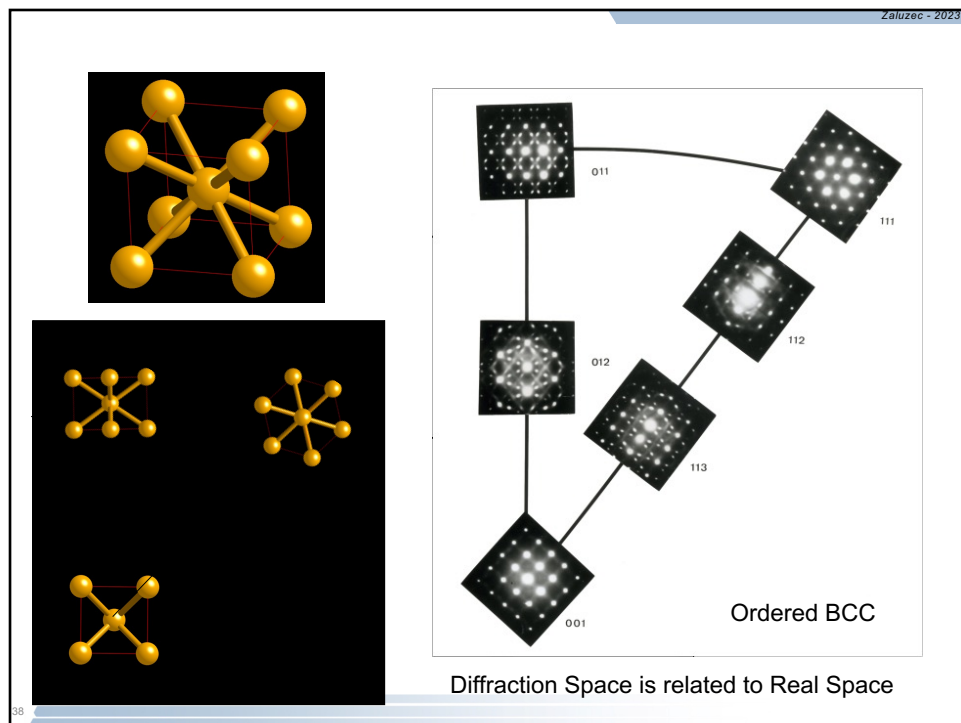
Table A4.1 Occurrence of reflections for the cubic crystal structures

Line no. $N = h^2 + k^2 + l^2$	hkl indices	$N^{1/2} = (h^2 + k^2 + l^2)^{1/2}$	f.c.c. diamond	Line no. $N = h^2 + k^2 + l^2$	hkl indices	$N^{1/2} = (h^2 + k^2 + l^2)^{1/2}$	b.c.c. f.c.c. diamond
1	100	1.00		33	522, 441	5.745	
2	110	1.414	×	34	530, 433	5.831	×
3	111	1.732	×	35	531	5.916	×
4	200	2.00	×	36	600, 442	6.00	×
5	210	2.236		37	610	6.083	
6	211	2.450	×	38	611, 532	6.164	×
7	—	—		39	—	—	
8	220	2.828	×	40	620	6.325	×
9	300, 221	3.00	×	41	621, 540, 443	6.403	×
10	310	3.162	×	42	541	6.481	×
11	311	3.317	×	43	533	6.557	×
12	222	3.464	×	44	622	6.633	×
13	320	3.606		45	630, 542	6.708	
14	321	3.742	×	46	631	6.782	×
15	—	—		47	—	—	
16	400	4.00	×	48	444	6.928	×
17	410, 322	4.123	×	49	700, 632	7.00	×
18	411, 330	4.243	×	50	710, 550, 543	7.071	×
19	331	4.359	×	51	711, 551	7.141	×
20	420	4.472	×	52	640	7.211	×
21	421	4.583		53	720, 641	7.280	
22	332	4.690	×	54	721, 633, 552	7.349	×
23	—	—		55	—	—	
24	422	4.899	×	56	642	7.483	×
25	500, 430	5.00	×	57	722, 544	7.550	×
26	510, 431	5.099	×	58	730	7.616	×
27	511, 333	5.196	×	59	731, 553	7.681	×
28	—	—		60	—	—	
29	520, 432	5.385		61	650, 643	7.810	
30	521	5.477	×	62	732, 651	7.874	×
31	—	—		63	—	—	
32	440	5.657	×	64	800	8.00	×

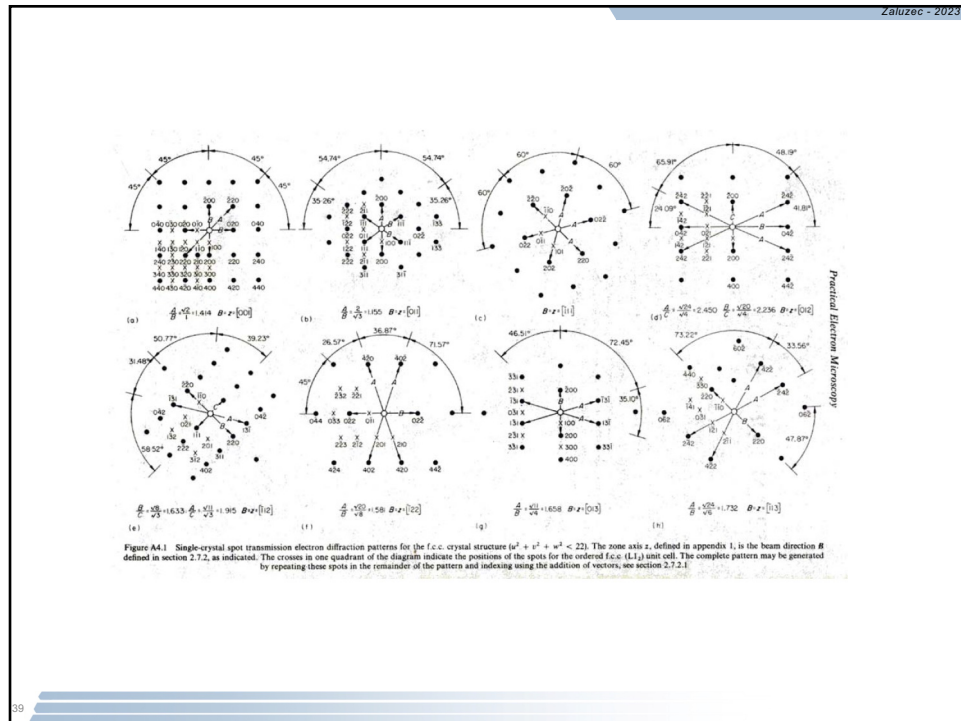
36



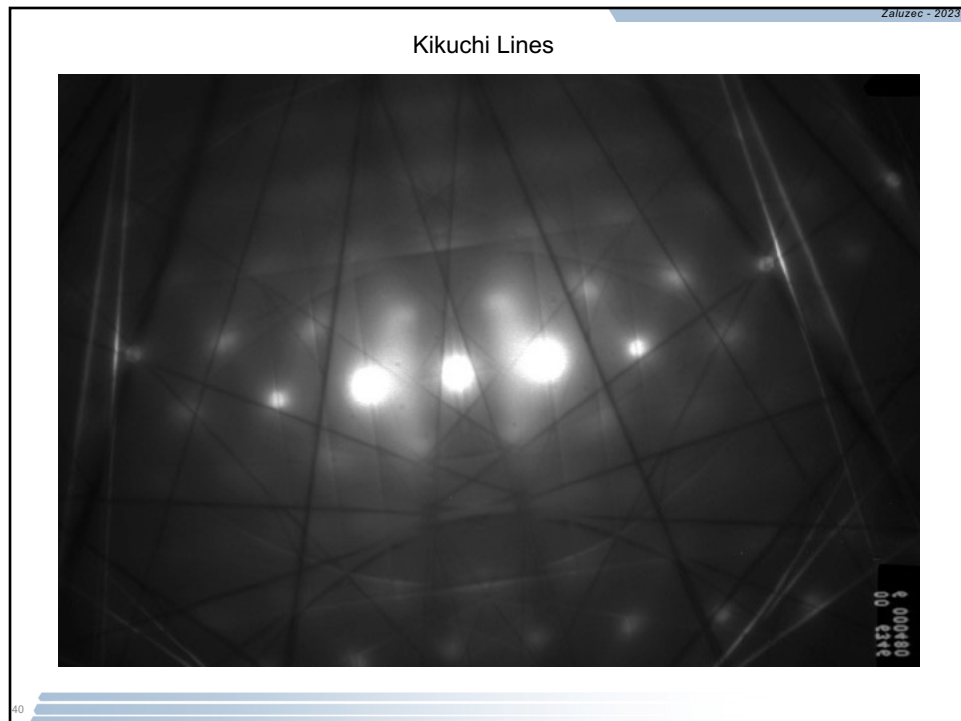
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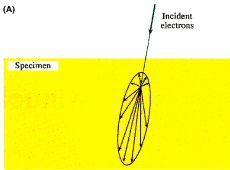


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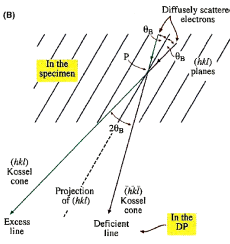
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Kikuchi Lines

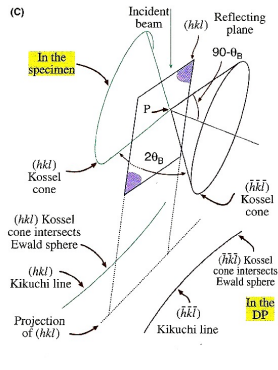
(A)

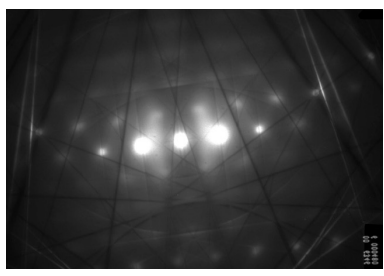
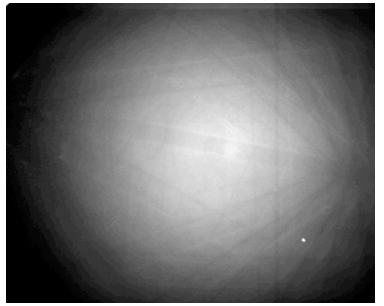


(B)



(C)



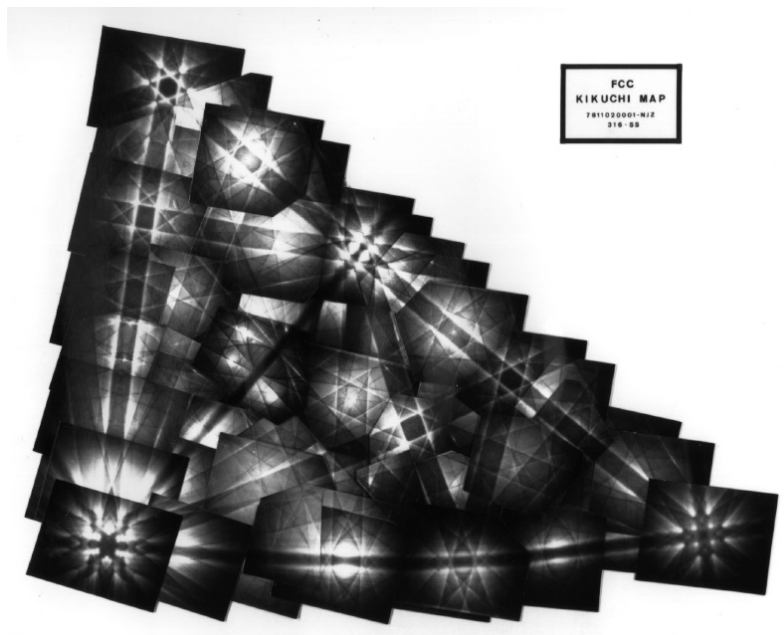



- Inelastically scattered electrons are Bragg Diffracted by lattice planes
- Range of orientation of scattered electrons gives rise to two cones of Diffraction
- Projection of the cones creates a "pair" of Kikuchi lines

41

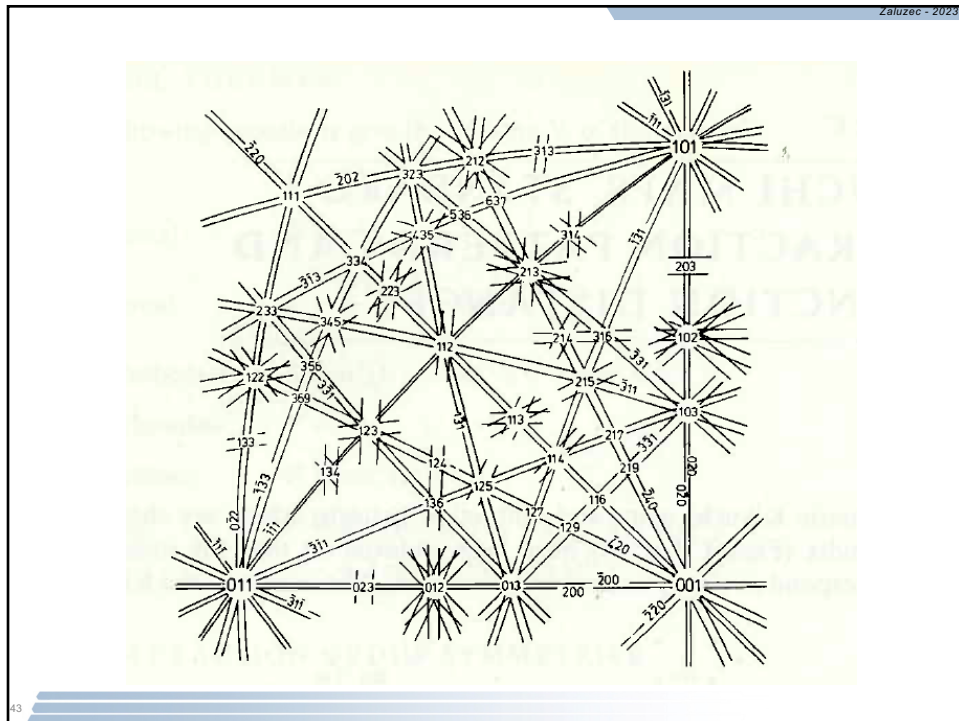
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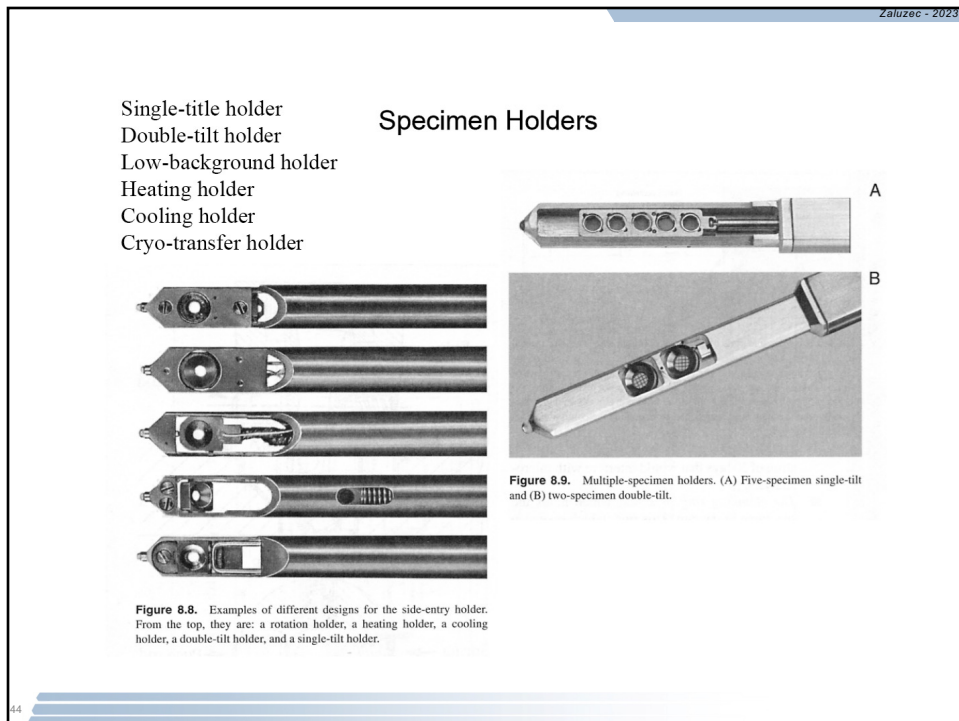


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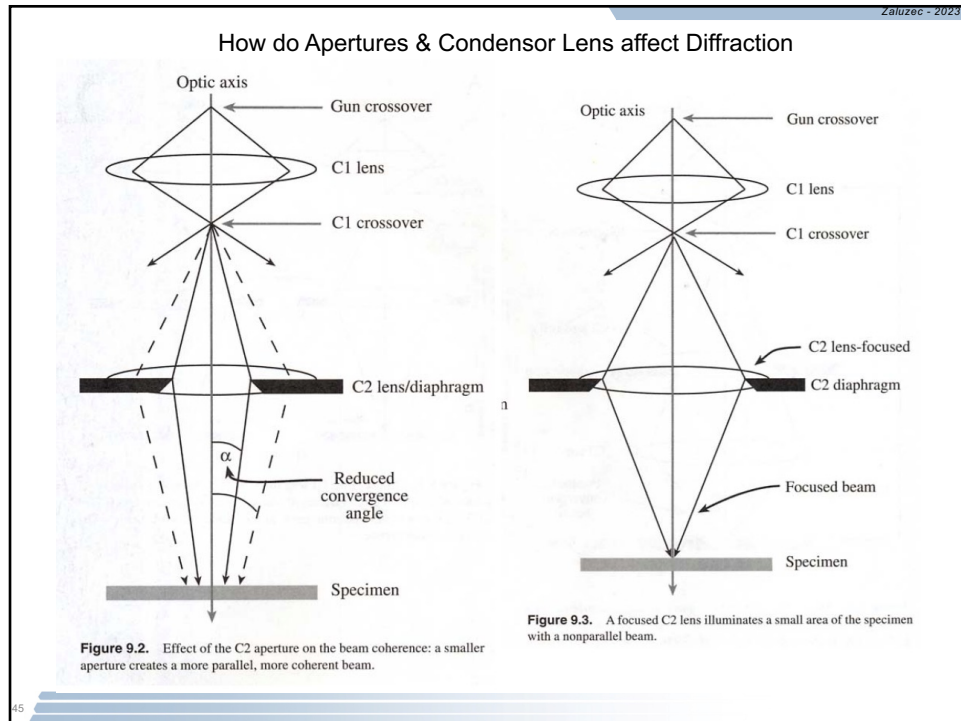
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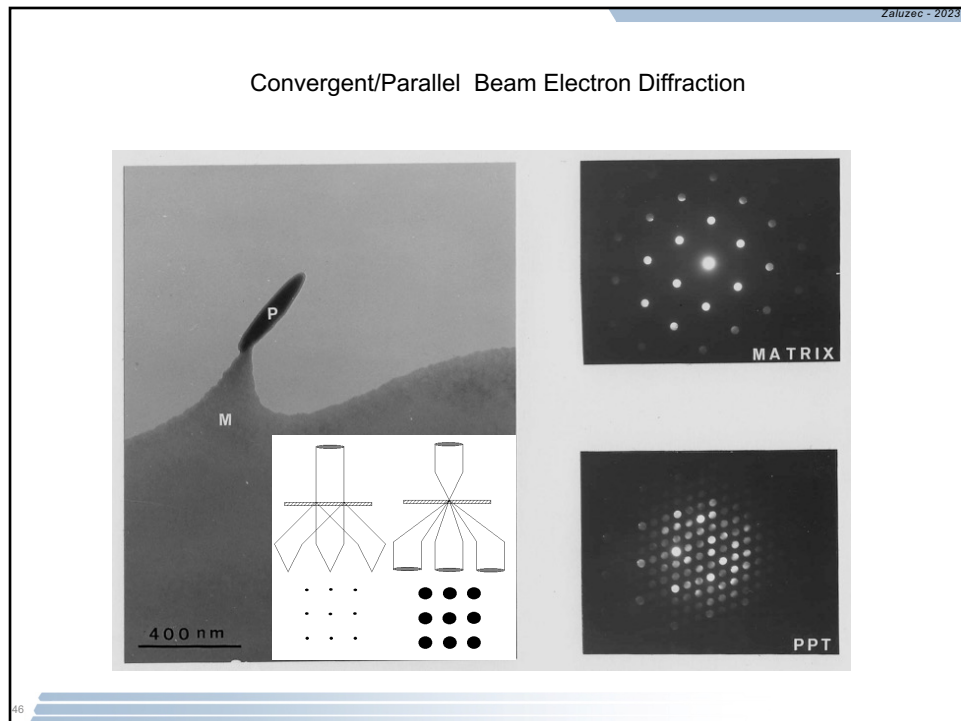
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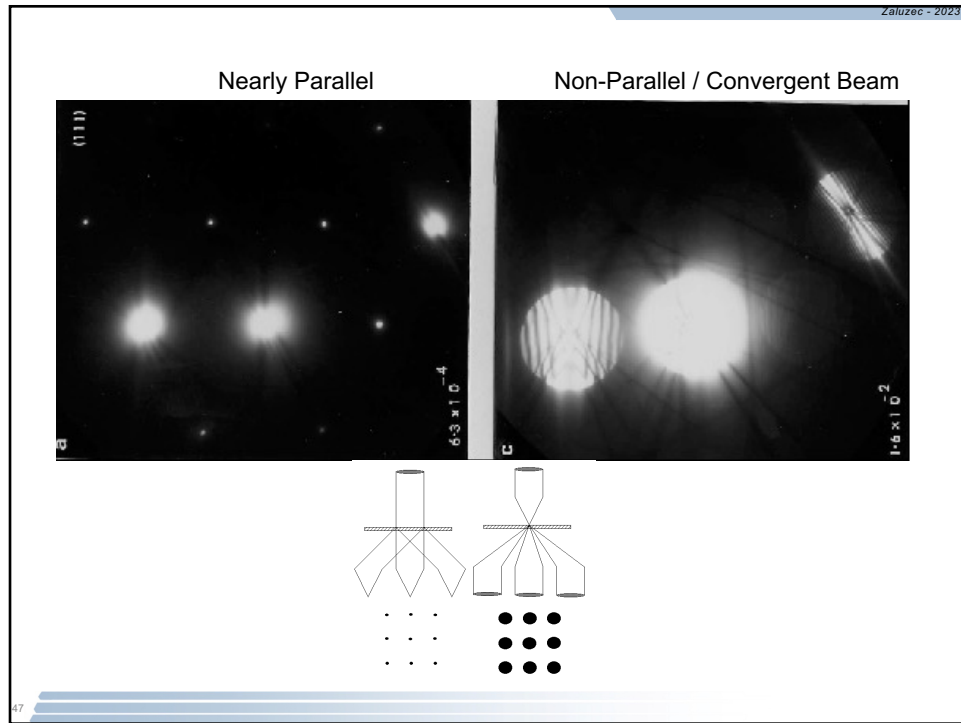
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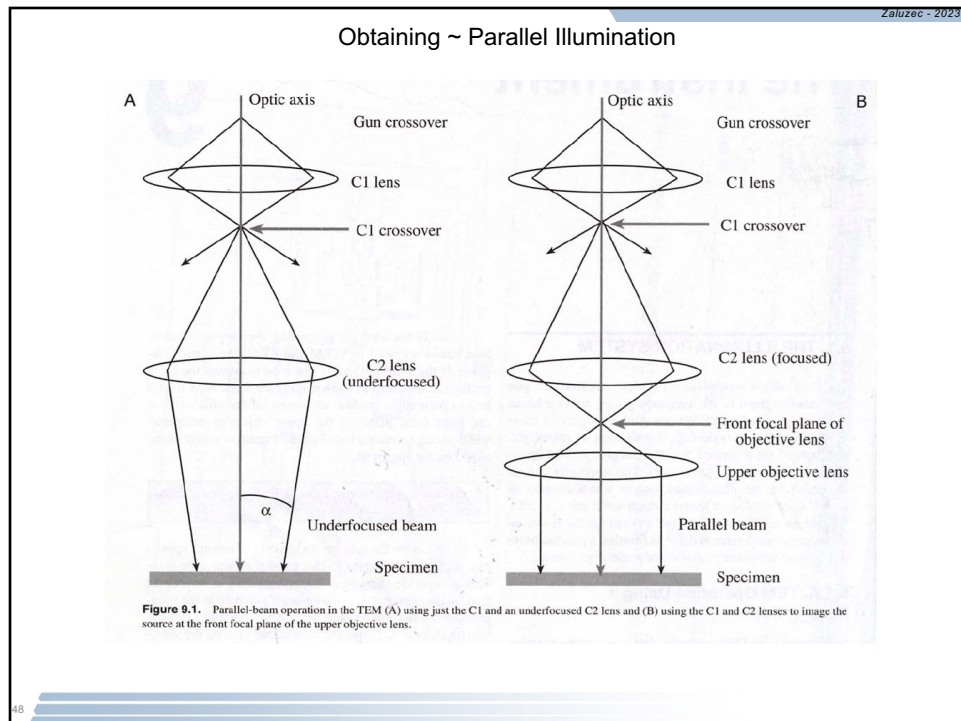
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Place an aperture in the Image Plane this defines an area back projected onto the specimen

Selected-Area Diffraction

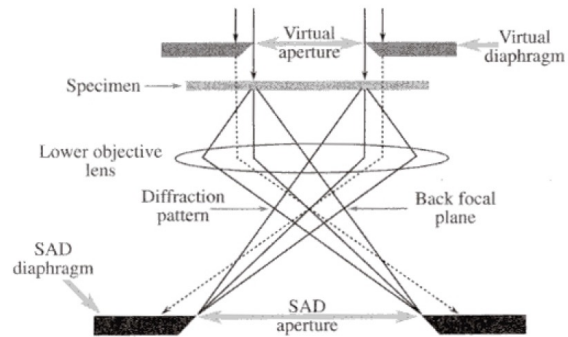
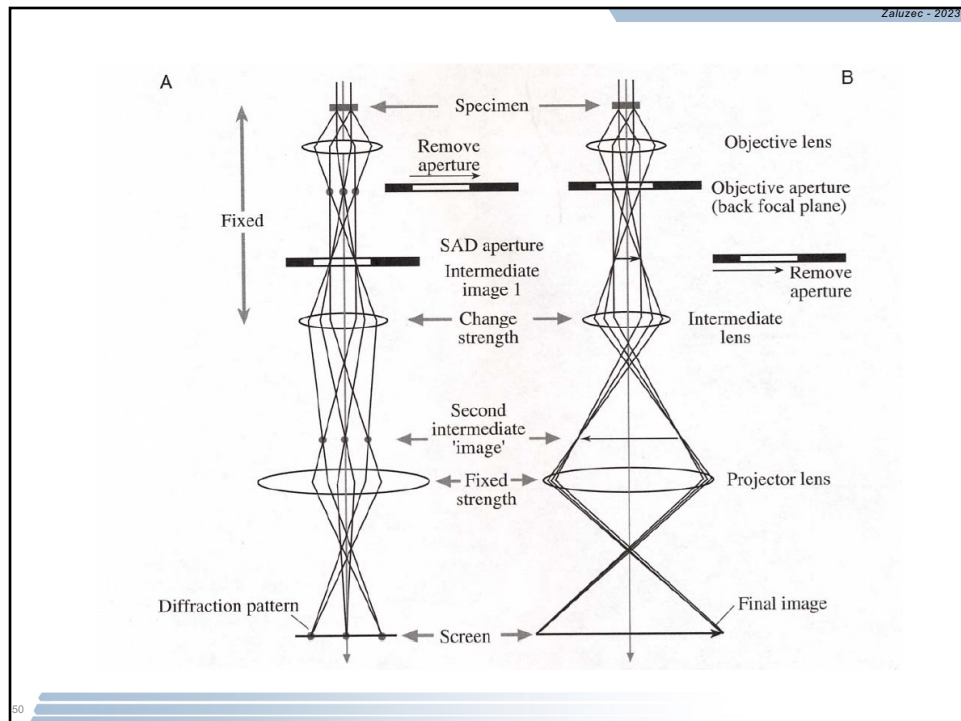


Figure 9.13. Ray diagram showing SAD pattern formation: the insertion of an aperture in the image plane results in the creation of a virtual aperture in the plane of the specimen. Only electrons falling inside the dimensions of the virtual aperture at the specimen will be allowed through into the imaging system. All other electrons will hit the SAD diaphragm.

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Why CBED? Why not SAD?

Limits of Conventional SAD

Conventional SAD uses an aperture to define the area from which the pattern is to be recorded. The aperture is placed in the image plane of the objective lens to create a virtual aperture in the specimen plane (Le Poole 1947). The spatial resolution in SAD is limited by both spherical aberration and the ability of the operator to focus the aperture of the and the image in the same plane. The error in area selection U is given by:

$$U = C_s(2\theta_B)^3 + D^2\theta_B$$

where: C_s = spherical aberration coefficient

θ_B = Bragg angle

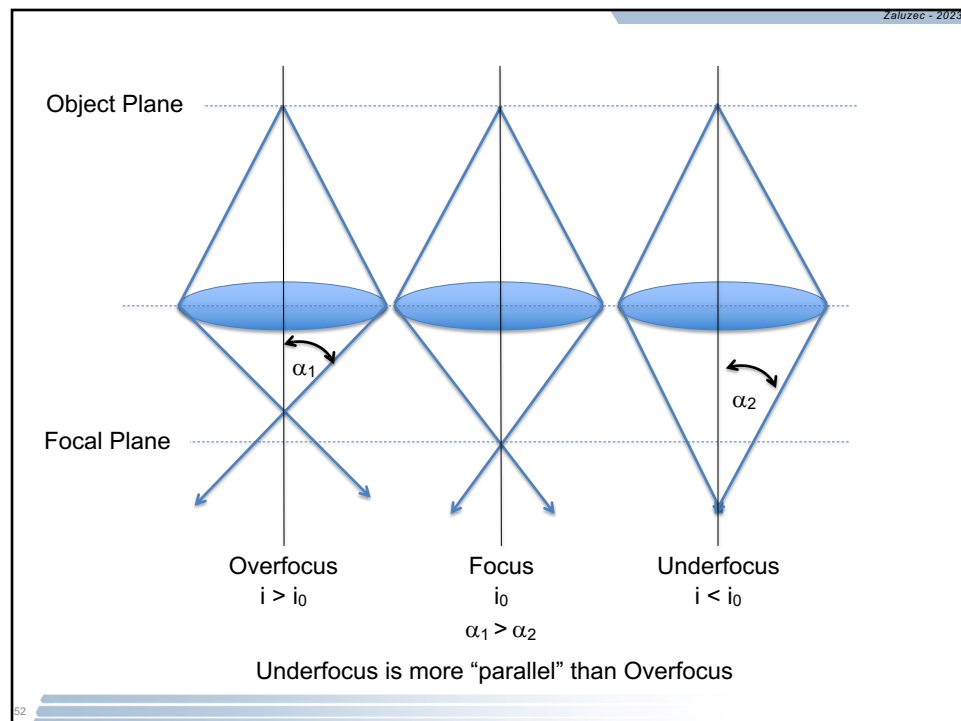
D = minimum focus step.

The result is that the theoretical lower limit of area selection is $\sim 0.5\mu\text{m}$ (in practice governed by aperture size).

J.B. Le Poole, Philips Tech. Rundsch 9 (1947) 33.

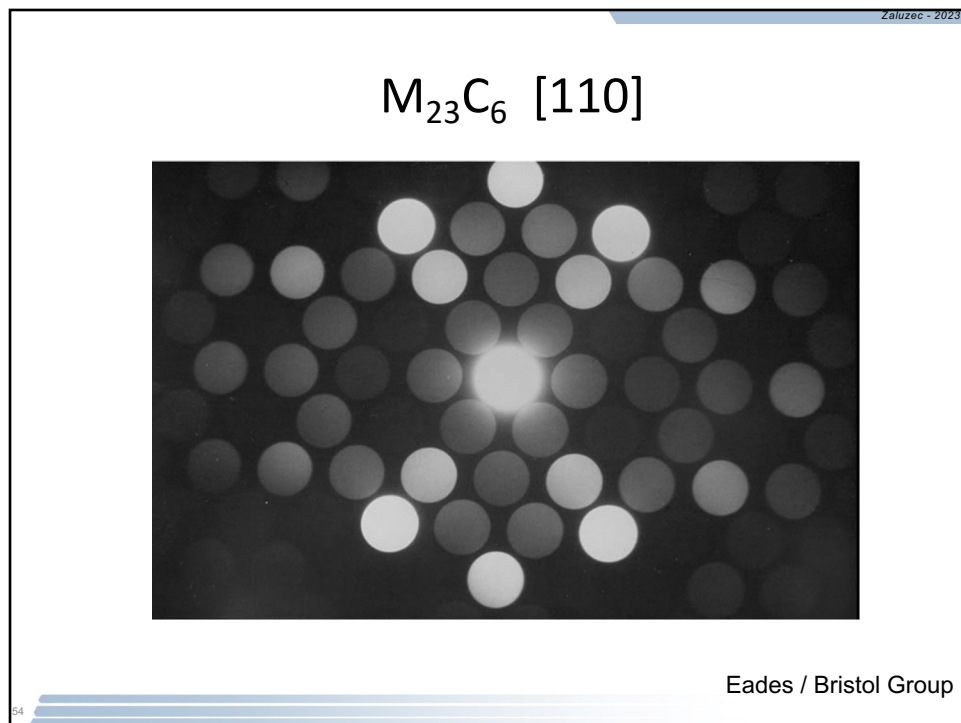
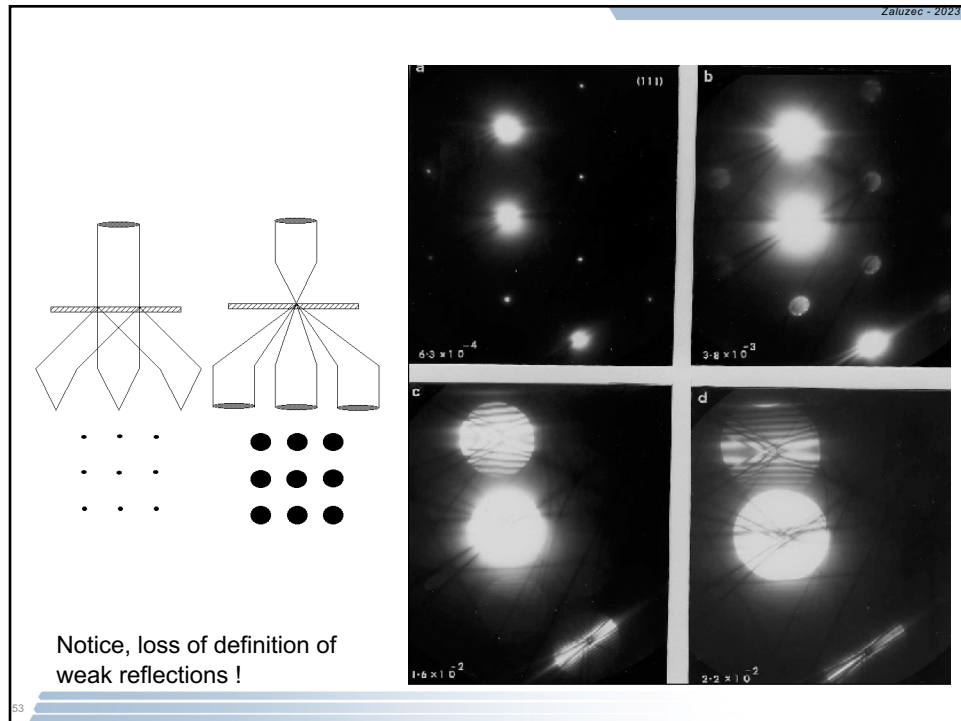
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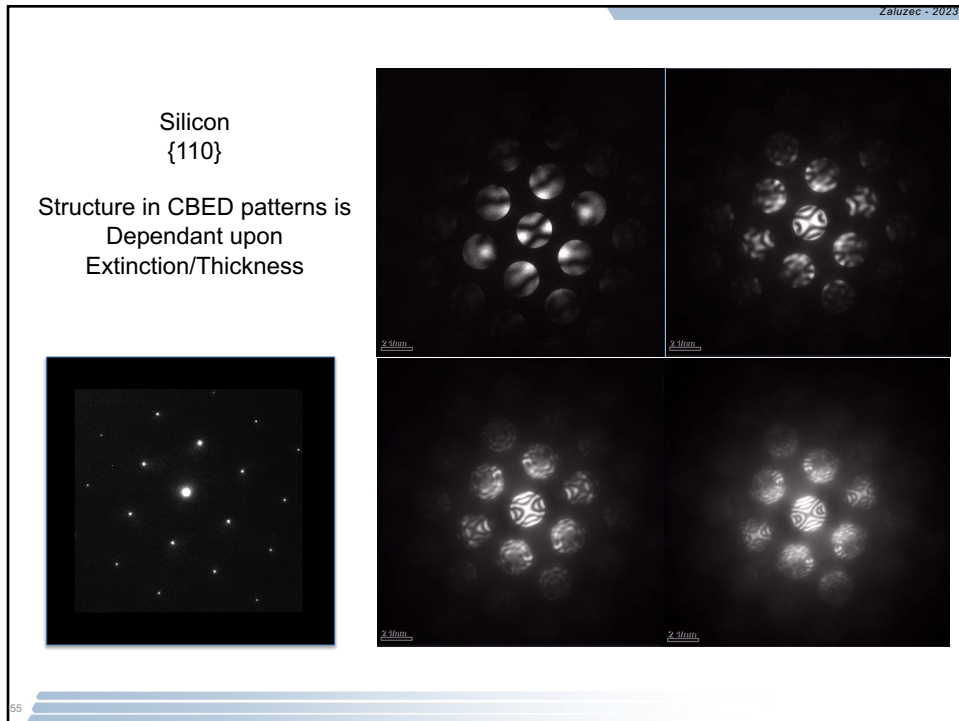
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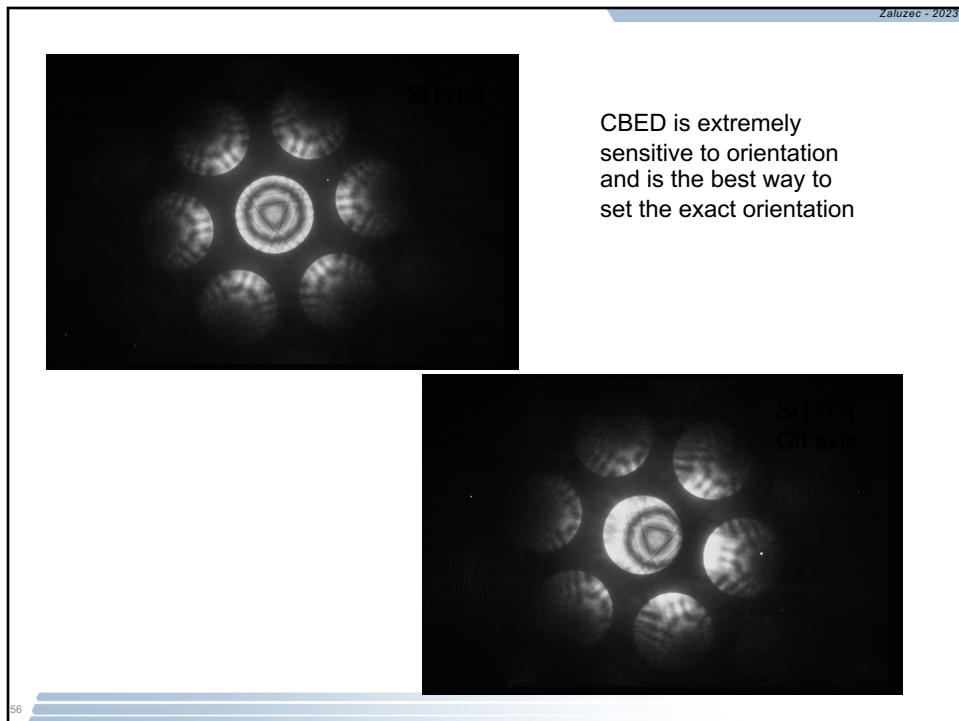
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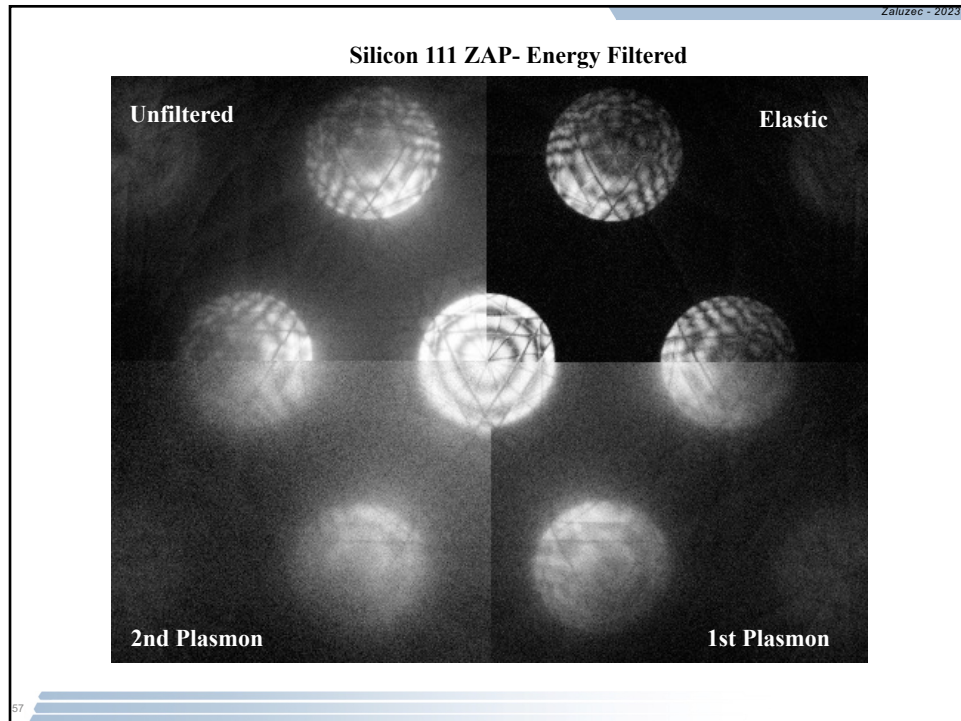




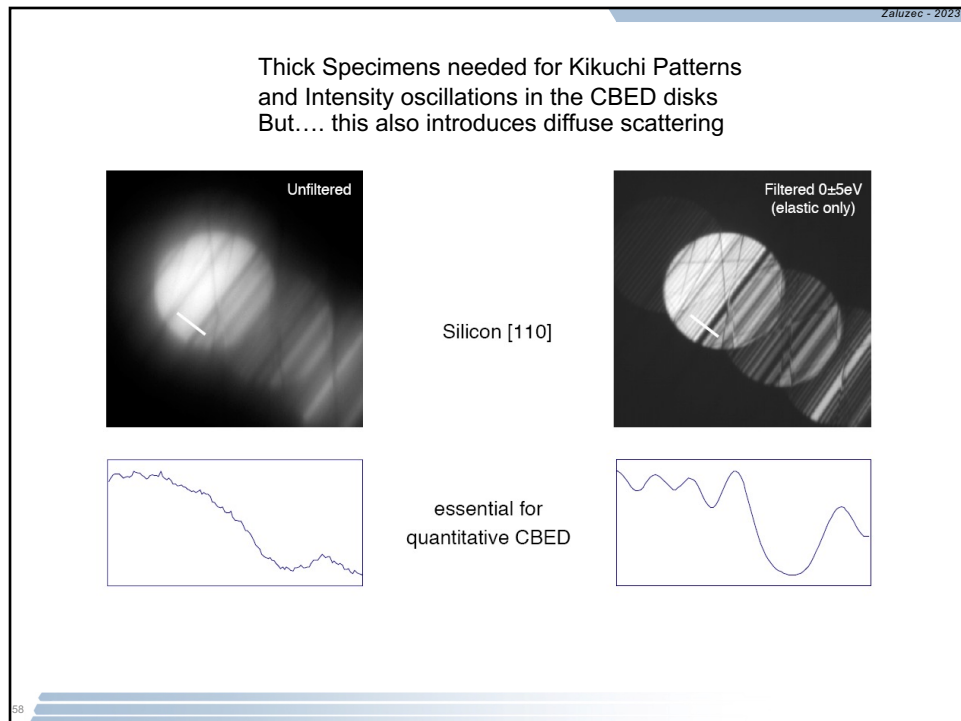
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Theoretical Considerations for Thickness Measurement

Convergent beam diffraction discs are maps of diffracted intensity as a function of incident wave angle and therefore have a direct correspondence to a rocking curve. In the two-beam approximation the rocking curve for the diffracted intensity $|\phi|^2$ is given by (Hirsch et al. 1965):

$$|\phi|^2 = \sin^2 \beta \pi \Delta k z \quad (1)$$

Where:

$$\beta = \tan^{-1} \left(\frac{1}{s \xi_g} \right) \quad \Delta k = \frac{\sqrt{1 + (s \xi_g)^2}}{\xi_g}$$

s is the deviation parameter, ξ_g is the extinction distance and z the foil thickness.

"Electron Microscopy of Thin Crystals",
Hirsch et al. (1965).

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Differentiation with respect to s reveals that the minima of $|\phi|^2$ in equation (1) obey the relationship:

$$\Delta k z = \text{integer} \quad (2a)$$

and the maxima obey the relationship:

$$\tan \pi \Delta k z = \pi \Delta k z \quad (2b)$$

Also, $s=0$ is always either a maximum or minimum. Kelly et al. (1975) expressed equation (2a) as:

$$\left(\frac{s_i}{n_k} \right)^2 = - \left(\frac{1}{\xi} \right)^2 \left(\frac{1}{n_k} \right)^2 + \left(\frac{1}{z} \right)^2 \quad (3a)$$

It is evident that a plot of $\left(\frac{s_i}{n_k} \right)^2$ against

$\left(\frac{1}{n_k} \right)^2$ in a two-beam condition yields a

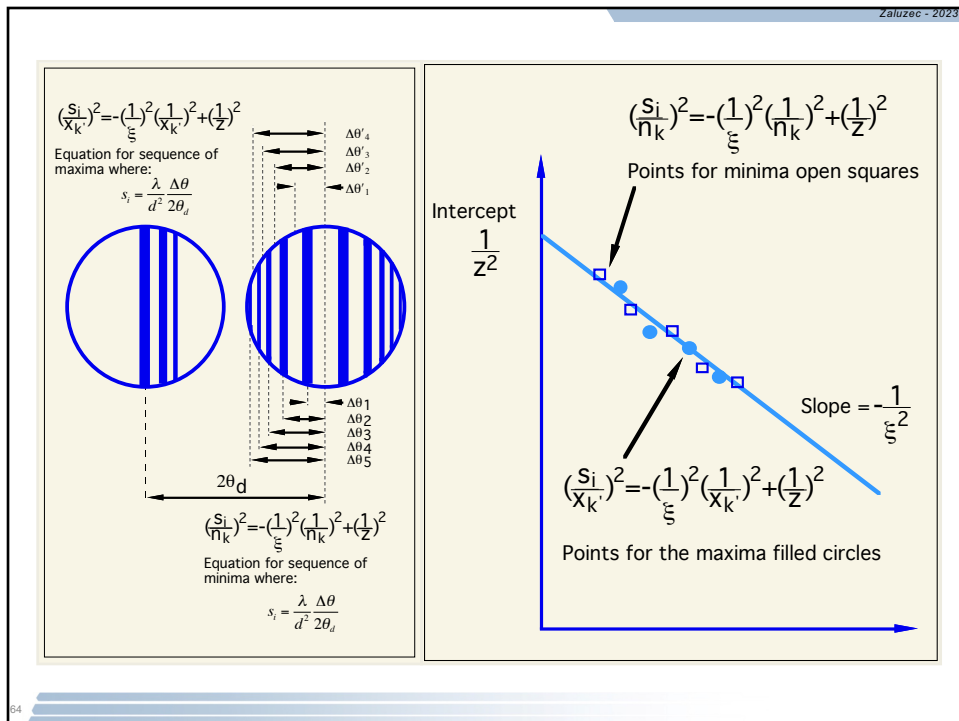
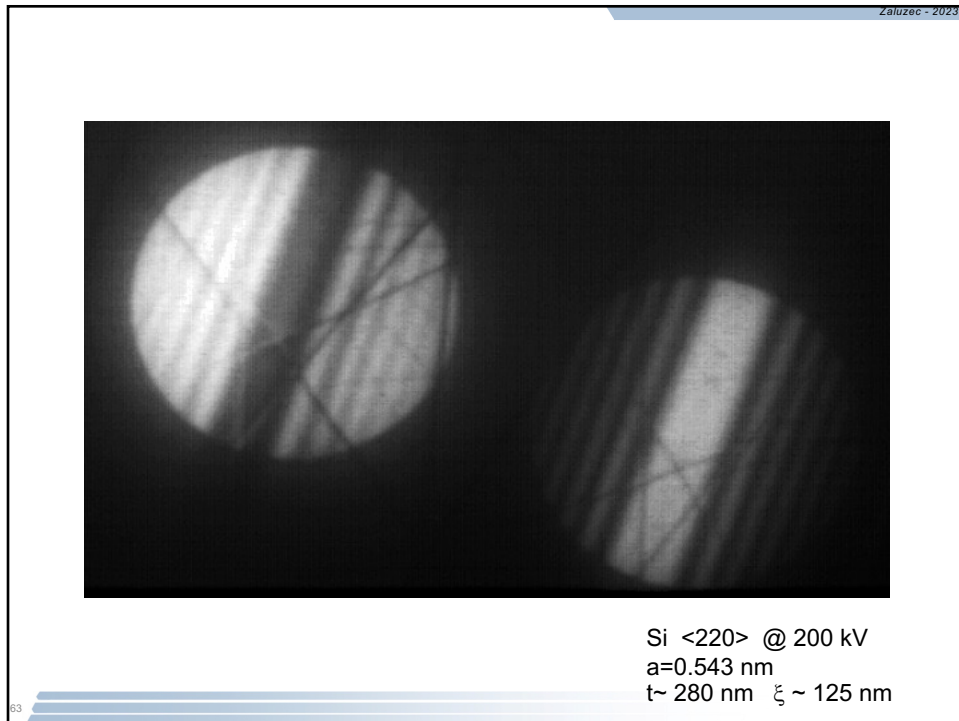
straight line with intercept $\left(\frac{1}{z} \right)^2$ and slope

of $\left(\frac{1}{\xi} \right)^2$. This is the basis of the CBED thickness measurement technique that is now well known.

P.M. Kelly et al.,
Phys. Stat. Sol.
(1975)A31, 771.

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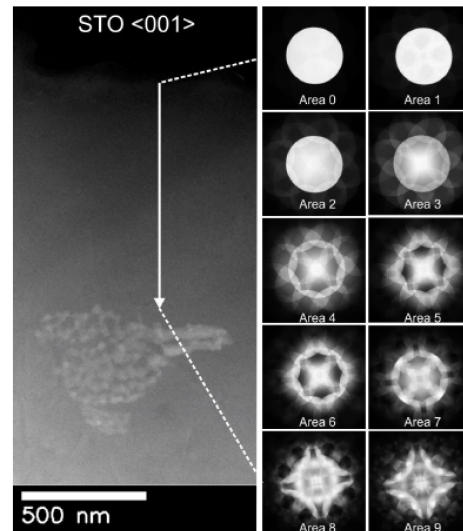


Position averaged CBED (PACBED) – crystal thickness measurement

- PACBED is a simple technique for crystal thickness measurement
- Scan over more than one unit cell
- Acquire a CBED pattern while doing so – this will average the pattern over an entire unit cell
- This replicates a large probe which averages over many unit cells – yet retains the overlaps from intermediate convergence angle
- Energy filter if you can, this will make things sharper

*Generally done in STEM mode
as you scan over an area*

Le Beau et al, Ultramicroscopy 110, p118 (2010)

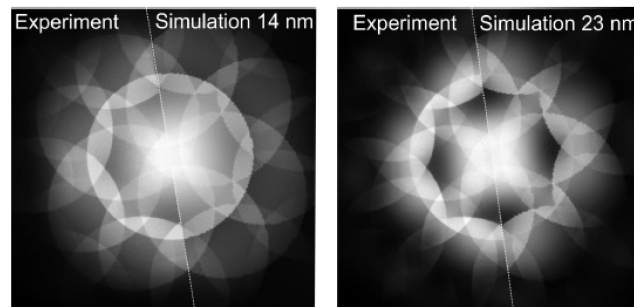


A specimen of STO – and a series of experimental PACBED patterns from increasing thickness

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Semi-quantitative measurement of thickness

- Compare simulated patterns with experimental patterns via visual inspection
- Simulation of incoherent patterns is fast (100 DP's can take a few minutes)
- Good packages for this include JEMS and μ STEM*
- Accuracy is on the order of 1 nm or so for intermediate thicknesses



Comparison between two experimental and simulated PACBED patterns

* <http://tcmp.ph.unimelb.edu.au/mustem/muSTEM.html>

But only measures Crystalline Thickness

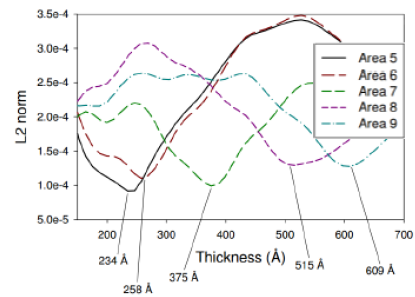
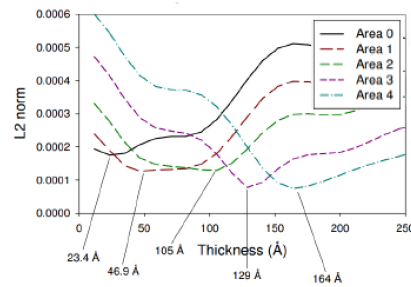
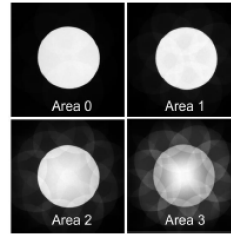
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Quantitative measure of crystal thickness...

Matching PACBED's by eye becomes challenging for thin crystals

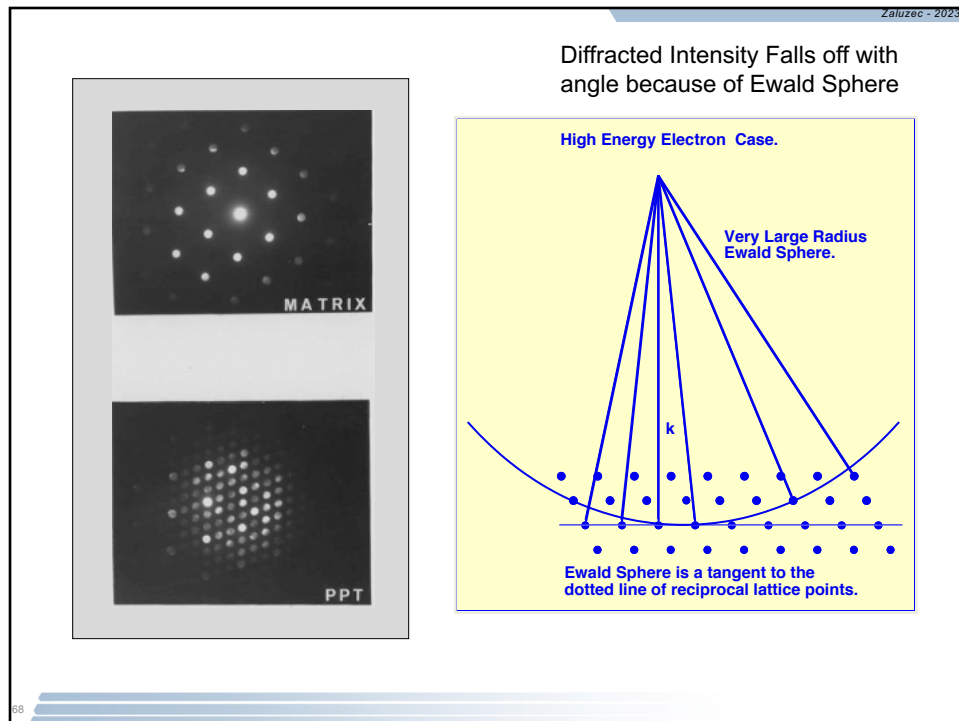
Currently matches are purely subjective

Least squares match (L2 norm) made to simulated PACBED patterns allows objective, repeatable thickness determination

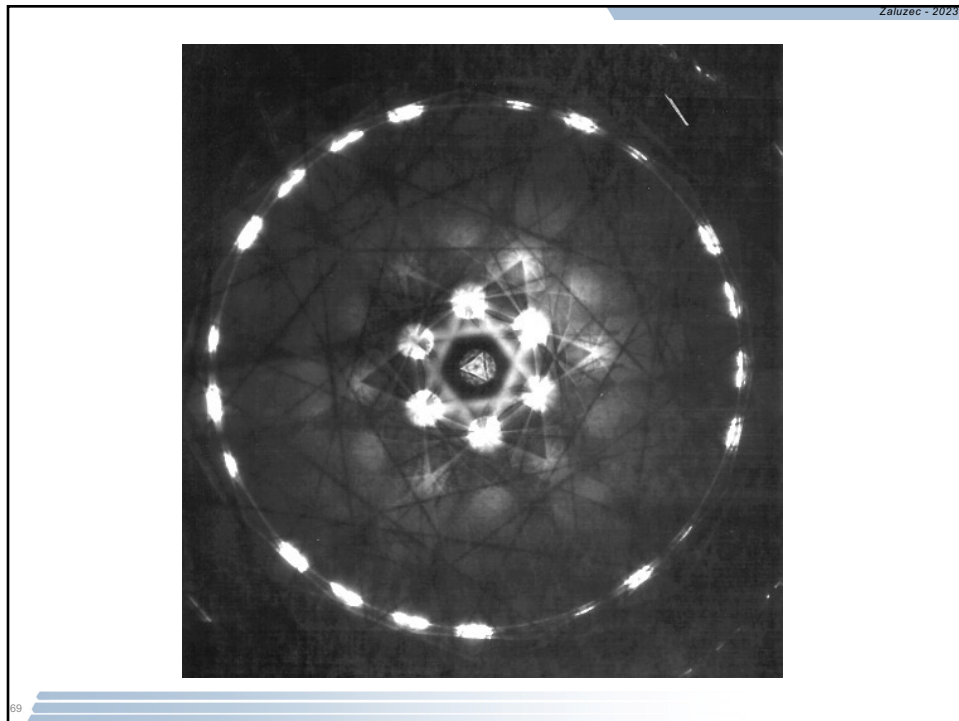


L2 norm match to simulations of experimental PACBED patterns, curve minima indicates most likely match

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Laue Zones

- At a zone-axis orientation, the reflections in the diffraction pattern break up into zones called Laue zones
- The central zone is called the zero-order Laue zone
- The first ring is called the first-order Laue zone - and so on
- The first-order, second-order, third order (and so on) are known collectively as the higher-order Laue zones

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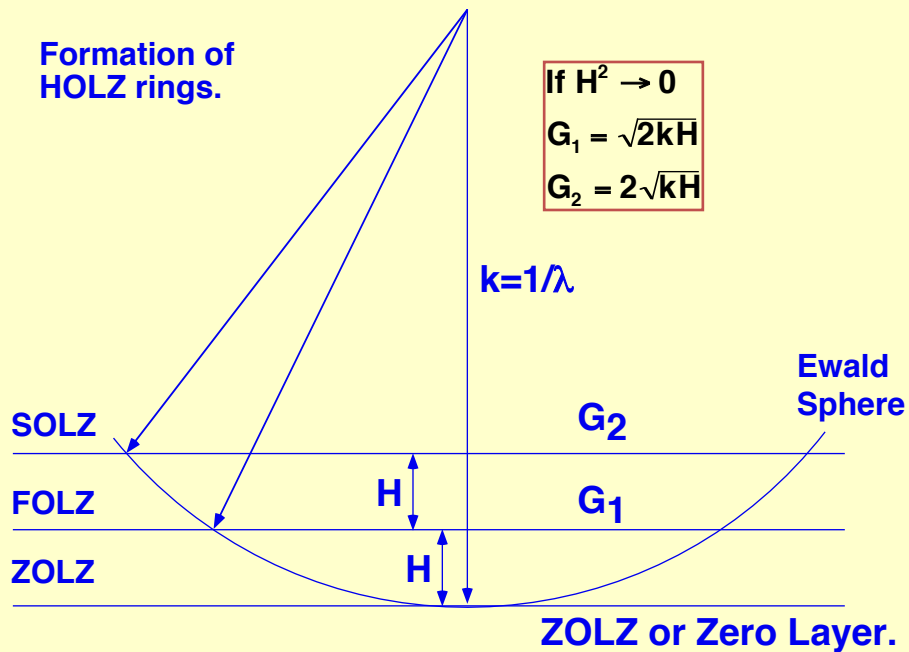
HOLZ

- HOLZ is the acronym for higher-order Laue zone
- The rings of reflections outside the central, zero-order Laue zone are the HOLZ
- Because the narrow, dark, straight lines in the bright field disc are associated with diffraction into a HOLZ reflection, they are known as HOLZ lines
- Do not confuse HOLZ with HOLZ lines

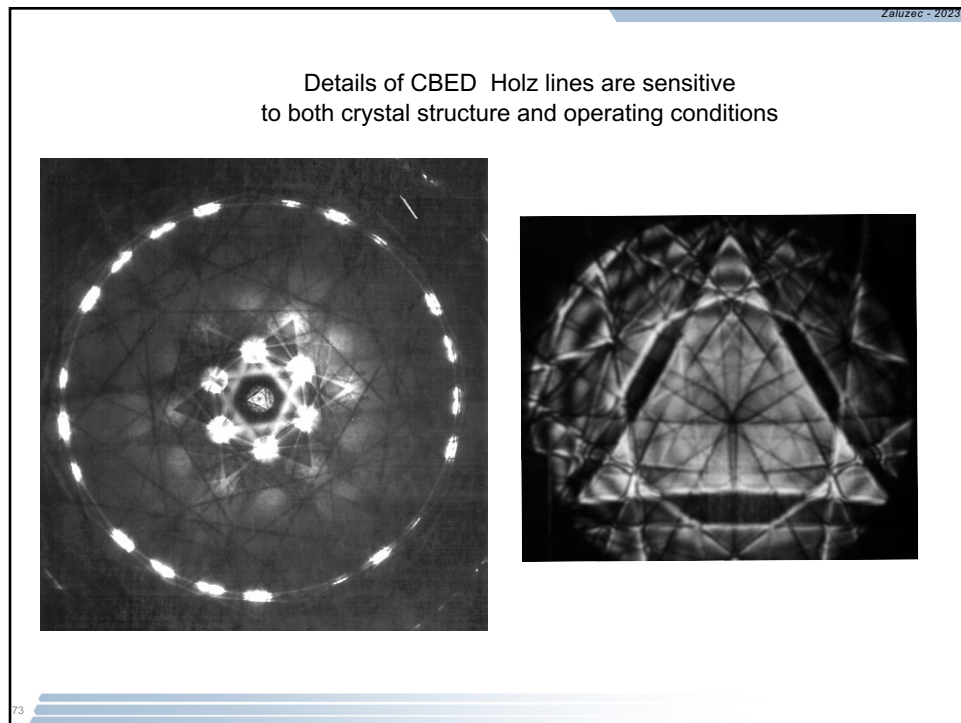
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Formation of
HOLZ rings.

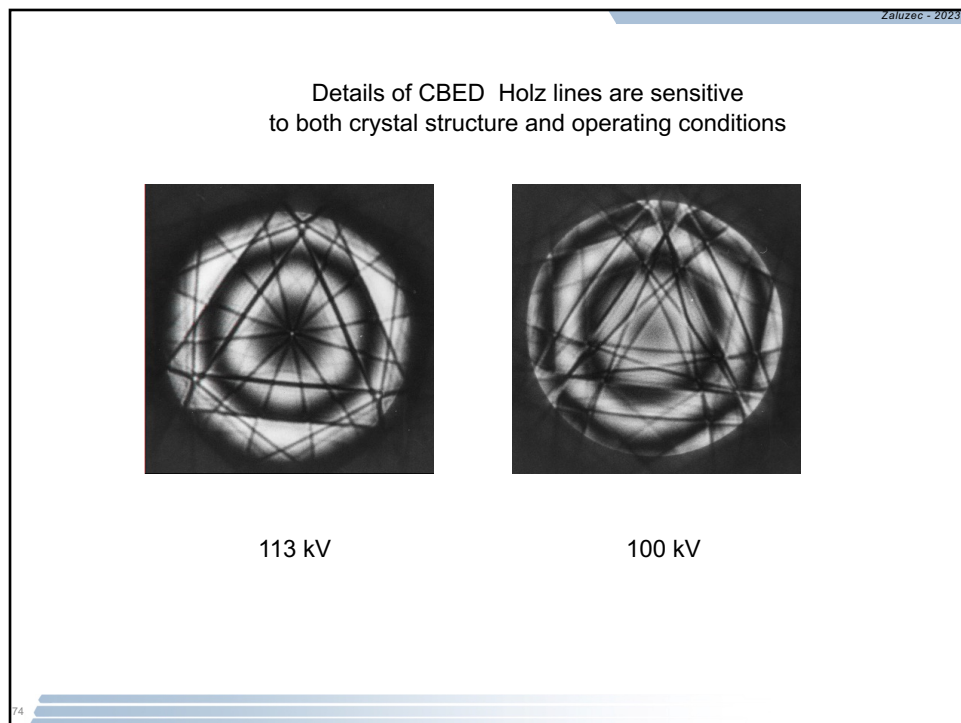
$$\begin{aligned} \text{If } H^2 &\rightarrow 0 \\ G_1 &= \sqrt{2kH} \\ G_2 &= 2\sqrt{kH} \end{aligned}$$



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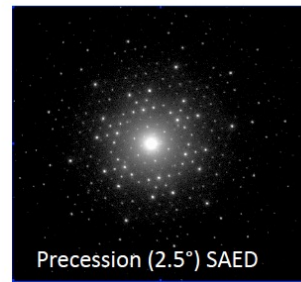
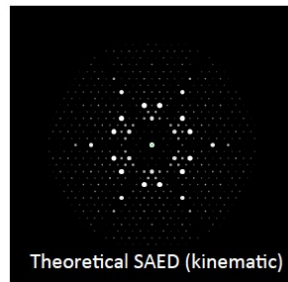
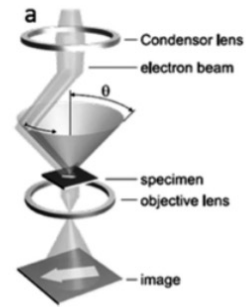
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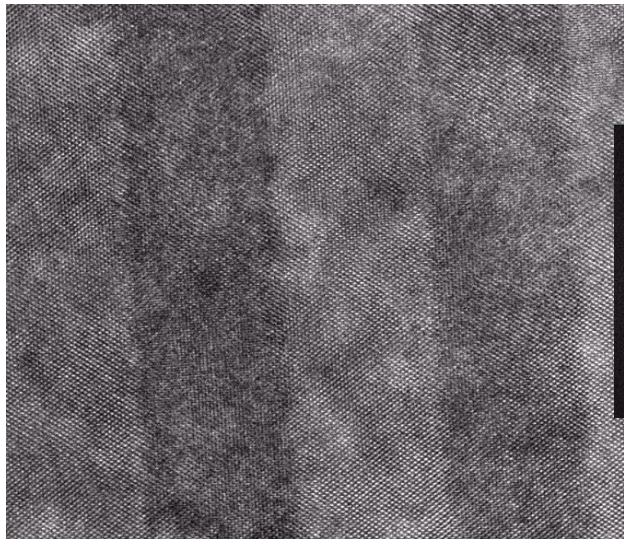
Precession diffraction

- SAED patterns are difficult for automated indexing
 - Dynamical diffraction
- Precession patterns are more kinematical
 - Can use diffraction intensities.
- Deflection coils used to tilt and rotate the beam in a cone with apex in the sample
 - Precession angle from $\sim 0.5^\circ$ to 3°

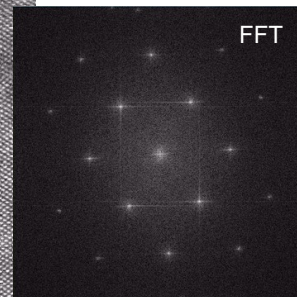


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*FFT of an Image is **NOT** a Diffraction Pattern !!!*



Si/Ge MultiLayer



It is a Power Spectrum
of the frequencies in
the Image

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